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Analysis of the electronic structure of the primary electron donor of photosystem I of *Spirodela oligorrhiza* by photochemically induced dynamic nuclear polarization (photo-CIDNP) solid-state nuclear magnetic resonance (NMR)

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1. Determination of the level of isotope labeling in Synechocystis sp. PCC 6803 and duckweed by LC-MS.

Table S1.1. LCMS peak intensities and calculated $^{L}I_t$ and P_n values for unlabeled and labeled Chl *a* isolated from *Synechocystis* sp. PCC 6803

m/z	^{U}I	$^{U}I_{\%}$	^{L}I	$^{L}I_t$	P_n
893.5	429304.0	1	190318.8	190318.8	0.545
894.5	288949.2	0.67	245637.4	117540.6	0.336
895.5	73242.9	0.17	147672.0	36089.6	0.103
896.5	26567.0	0.06	56997.8	876.1	0.003
897.5	7339.1	0.02	16776.9	2756.2	0.008
898.5	1579.5	0.00	5400.6	1162.7	0.003
899.5	0.0	0.00	4324.3	782.6	0.002
900.5	0.0	0.00	0.0	0.0	0.000
901.5	0.0	0.00	0.0	0.0	0.000

Table S1.2. LCMS peak intensities and calculated $^{L}I_t$ and P_n values for unlabeled and labeled Chl *a* isolated from duckweed

m/z	^{U}I	$^{U}I_{\%}$	^{L}I	$^{L}I_t$	P_n
893.5	161481.0	1	118245.9	118245.9	0.289
894.5	120301.9	0.74	193086.8	104994.7	0.257
895.5	68342.5	0.42	173475.7	45211.2	0.111
896.5	9208.8	0.17	121856.1	36994.8	0.090
897.5	3323.8	0.02	75276.7	22594.0	0.055
898.5	1277.0	0.00	41455.4	6387.8	0.016
899.5	633.7	0.00	26502.6	10071.8	0.025
900.5	0.0	0.00	28208.2	16712.9	0.041
901.5	0.0	0.00	64674.5	47596.7	0.116

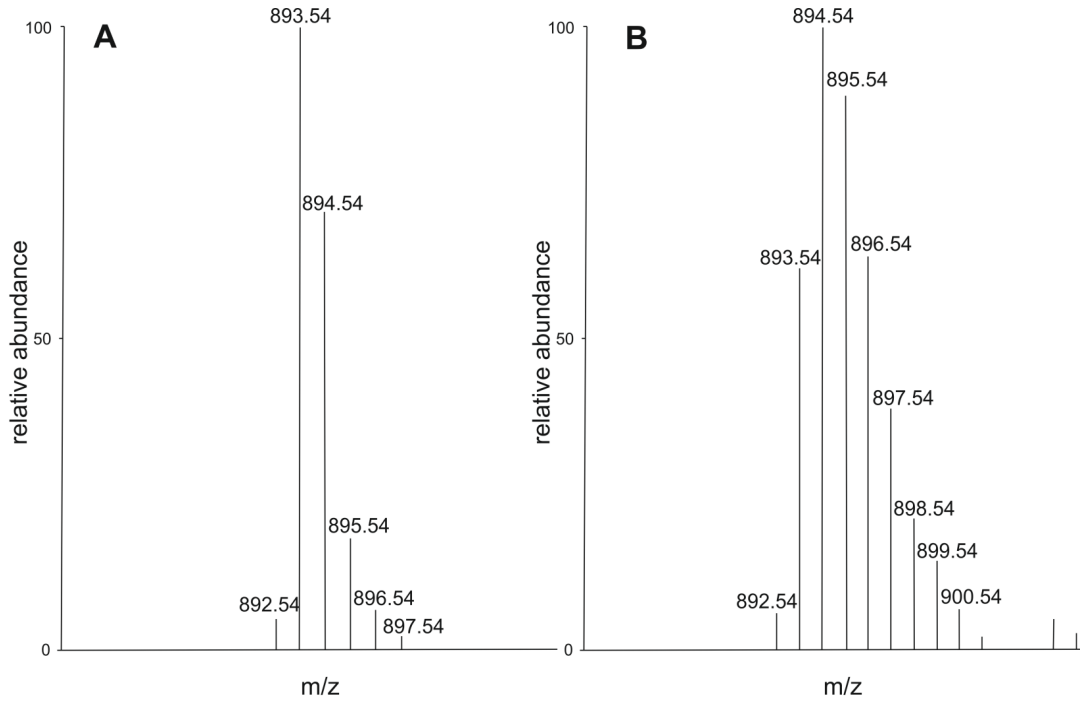

Figure S1.1: ^{13}C isotope incorporation determined by LCMS for Chl *a* isolated from duckweed leaves grown on unlabeled substrate (A) and with the ^{13}C 4-ALA precursor in the medium (B).

2 Computational Details

2.1 Model Setup

The desired models were extracted from the crystal structure of Photosystem I in plants (PDB entry 2WSC (Amunts et al., 2010), provided by the Protein Data Bank (Berman et al., 2000). The protein and non-protein part of the models was processed separately. The *Reduce* program (Word et al., 1999) was used to add missing hydrogen atoms of co-factors. This procedure was not accurate enough to add hydrogen atoms of water molecules, why they were added manually with *Avogadro* (Hanwell et al., 2012). Missing hydrogens of the protein were added using *CHARMM22* topology files (MacKerell et al., 1998; MacKerell et al., 2004) with the *Automatic PSF Builder* within the *VMD* (Humphrey et al., 1996) program. Additionally, cut protein bonds were saturated with neutral groups -C(O)CH_3 and -NH_2 as N- and C-terminus, respectively. The values of bond distances for the C- and N- termini were manually adjusted. The distances of corresponding groups in the crystal structure were added to a modified topology file (see Fig. S2.1). Binding pocket models of PSI were created by specifying radii of 3.2 and 3.4 Å around each atom of the co-factor of interest. All surrounding co-factors, water molecules, and amino acid residues with at least one atom within these radii were included explicitly into the models. Addition of missing hydrogens and cut bond saturation were conducted in a similar fashion as for smaller models considered in this work (see above). The binding pocket models are indicated by abbreviations corresponding to the chosen radii “r32” and ”r34”, while the isolated co-factor models are abbreviated “iso”.

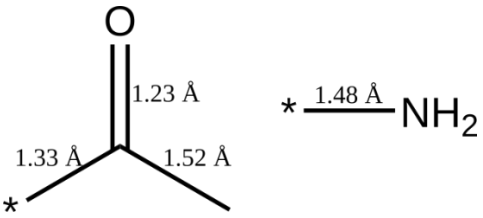

Figure S2.1: Bond distances for the N- and C-termini.

2.2 Geometry setup

To assess the quality of the calculated chemical shifts, the structure of P_A was extracted from the crystal structure, truncated and processed according to Sec. 2.1. The molecular structure was then fully optimized using the *Turbomole v7.4.1* (Ahlrichs et al., 1989; TURBOMOLE, 2019) program package. This model will be called ”isoOpt” in the following. For the optimization the def2-SVP basis set (Schäfer et al., 1992) and PBE0 functional (Perdew et al., 1996a; Adamo and Barone, 1999) were used. Additionally, the resolution-of-the-identity approximation in conjunction with the auxiliary Coulomb fitting basis (Schäfer et al., 1992; Schäfer et al., 1994) was enabled. Dispersion corrections were taken into account with the D3 correction with Becke–Johnson damping (Grimme et al., 2010; Grimme et al., 2011). Secondly, the structure of P_A was extracted from the crystal structure and optimized using the DFTB3 (Gaus et al., 2012) method, which is described in detail in the following. This model will be called “Pa extract”.

For all other geometry optimizations, the DFTB3 (Gaus et al., 2012) method within the *AMS-DFTB* module from the *ADF* 2019 package (Amsterdam; Velde et al., 2001) was used. The "Third-Order Parametrization for Organic and Biological Systems" (3ob) (Gaus et al., 2013; Kubillus et al., 2015) parameters from the corresponding Slater–Koster file were used. The optimization was carried out within two steps, where in the first step the coordinates corresponding to hydrogen atoms were optimized, while all other nuclear coordinates were kept fix. In the second step for each Chl co-factor the atoms C-3^1, C-3^2 and their attached hydrogen atoms were optimized while all other coordinates were kept fix. This step was done, to encounter the poor quality of this type of double bond in the crystal structure, where atoms C-3, C-3^1 and C-3^2 are oriented linearly. The general two-step procedure ensured to keep the relative arrangement of co-factors and environment residues, while the co-factor's geometry is partly relaxed in the presence of the protein pocket.

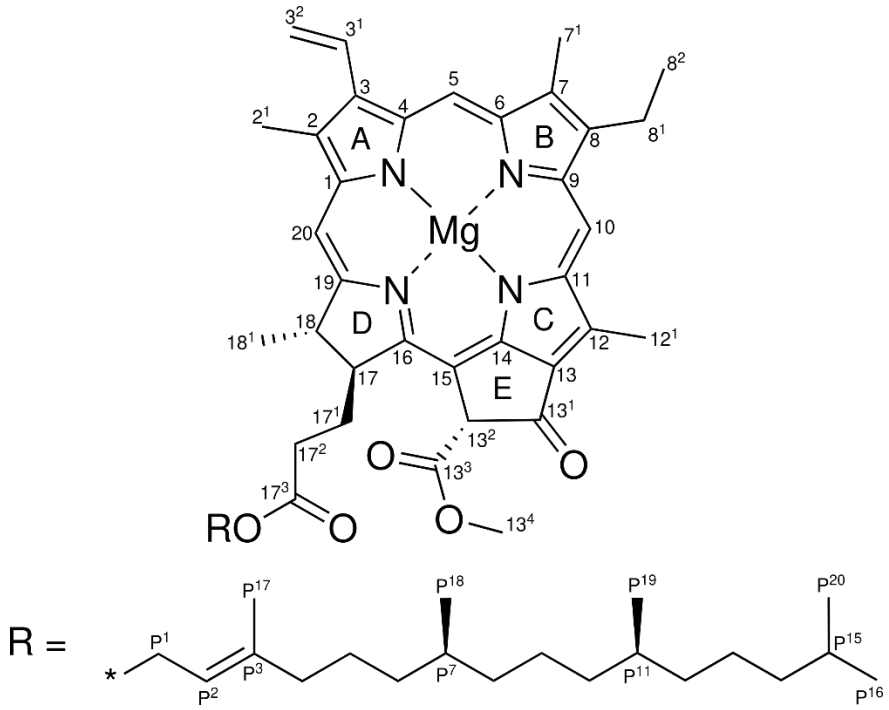

Figure S2.2: Lewis structure of Chl *a*. The atomic numbering is given according to IUPAC. Nitrogens N-I, N-II, N-II and N-IV are localized in pyrrole rings A, B, C and D, respectively.

2.3 NMR Calculations

For the "iso" models ^{15}N and ^{13}C nuclear magnetic shieldings were calculated using the KT2 (Keal and Tozer, 2003) exchange-correlation functional and a TZP (Perdew et al., 1992) basis set from the *ADF* 2019 package (Amsterdam; Velde et al., 2001) library. The numerical quality of the density fit and grid construction procedures were set to "good". For binding pocket models, labeled as "r32" or "r34", calculations were carried out within a subsystem DFT approach (Jacob and Visscher, 2006; Jacob and Neugebauer, 2014; Wesolowski et al., 2015) using the TZP (van Lenthe and Baerends, 2003) basis set and the PW91 (Perdew

and Wang, 1991; Perdew et al., 1992) exchange-correlation functional with the conjoint (Lee et al., 1991) kinetic-energy functional PW91k (Lembarki and Chermette, 1994). Mutual relaxations of subsystem densities were accounted for using 3 freeze-and-thaw (FaT) cycles (Wesolowski and Weber, 1996). ^{15}N chemical shifts were calculated with respect to the ammonia shieldings, while ^{13}C chemical shifts were calculated with respect to tetramethylsilane (TMS). Therefore, both molecules were optimized using the KT2 (Keal and Tozer, 2003) exchange-correlation functional and a TZP (Perdew et al., 1992) basis set from the *ADF* 2019 package (Amsterdam; Velde et al., 2001) library. ^{15}N as well as ^{13}C nuclear magnetic shieldings were calculated, where the chemical shift was calculated by $\delta_{\mathrm{i}} = \sigma_{ref} - \sigma_{\mathrm{i}}$, where σ_{ref} and σ_i are the chemical shielding of the reference and atom of interest, respectively. Ring current effects of other subsystems were considered by calculating nuclear independent chemical shifts (NICS) as it was done in Jacob and Visscher (2006).

For comparison of the NMR calculations of the "isoOpt" and "Pa extract" models the *ADF* program with the PBE functional (Perdew et al., 1996b) and triple-ζ-basis set was used.

2.4 Calculation of the Environment effect

To calculate the effect of the protein environment on the chemical shifts of the co-factors the following equation was applied,

$$\Delta\delta = \delta\left(\mathrm{CoF}_{\mathrm{A/B}}^{34}\right) - \delta\left(\mathrm{CoF}_{\mathrm{A/B}}^{\mathrm{iso}}\right). \quad (1)$$

In the equation above $\delta\left(\mathrm{CoF}_{\mathrm{A/B}}^{34}\right)$ is the chemical shift of a particular co-factor $\mathrm{CoF}_{\mathrm{A/B}}$ within a protein environment of 3.4 Å, where $\delta\left(\mathrm{CoF}_{\mathrm{A/B}}^{\mathrm{iso}}\right)$ is the chemical shift of a particular isolated cofactor. The difference $\Delta\delta$ is, thus, the effect of the protein environment on the chemical shifts of CoF.

2.5 Validation of the NMR Calculations

In this section results are shown that assess the quality of the calculated chemical shifts. Therefore, ^{15}N Chemical shifts obtained experimentally by Boxer et al. (1974) and by the authors of this work, which agree very well, are compared to shifts calculated with *ADF* for the "isoOpt" and "Pa extract" model. The data are shown in Fig. S2.3. The calculated chemical shifts are overestimated by about 10 to 20 ppm compared to the experimental value, but correctly predict the trend of the chemical shifts (see also Fig. S2.4). Additionally, the calculated shifts of N_{II} and N_{III} for "Pa extract" are in very good agreement with the experimental values, while N_I and N_{IV} deviate from the experimental results. The reader should note here, that the results for N_{IV} strongly deviate in all four shown lines (see Figs. S2.3 and S2.4).
The deviations in the calculated NMR shifts strongly depend on the geometry of the calculated models. The values calculated for "Pa extract" show good agreement with the experiment especially for nitrogen atoms 2 and 3, while the "isoOpt" values better represent the general trend of the shifts. This shows that a possible error source for the calculated NMR shifts arises from the use of the static crystal structure rather than averaging over conformations accessible during the protein dynamics. This could be assessed by performing

a short molecular dynamics simulation and calculating NMR shifts for an ensemble of structures, which is, however, beyond the scope of this work.

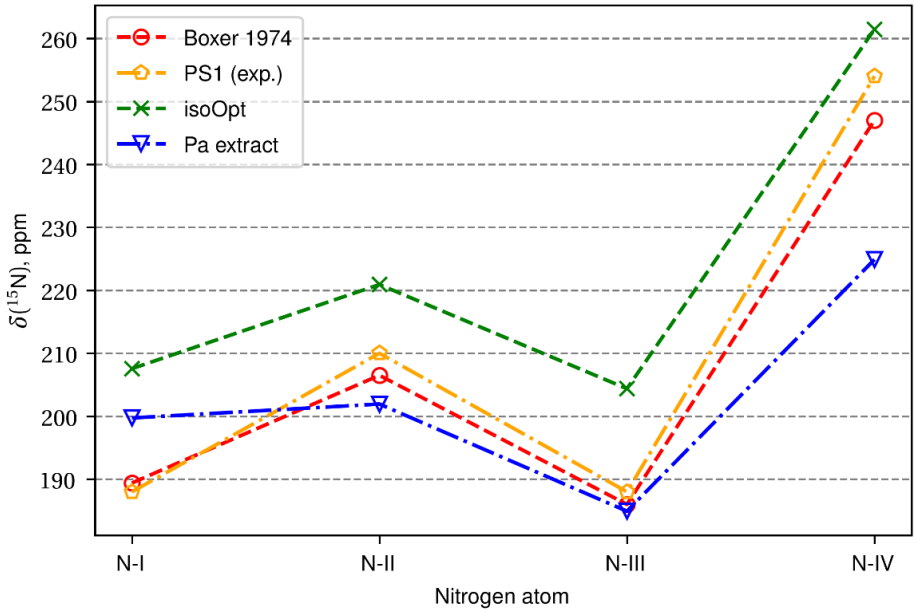

Figure S2.3: Comparison of ^{15}N Chemical shifts obtained experimentally by Boxer et. al. [31] (red) and the authors of this work (orange) with those calculated with ADF for the "isoOpt" (green) and "Pa extract" (blue) model.

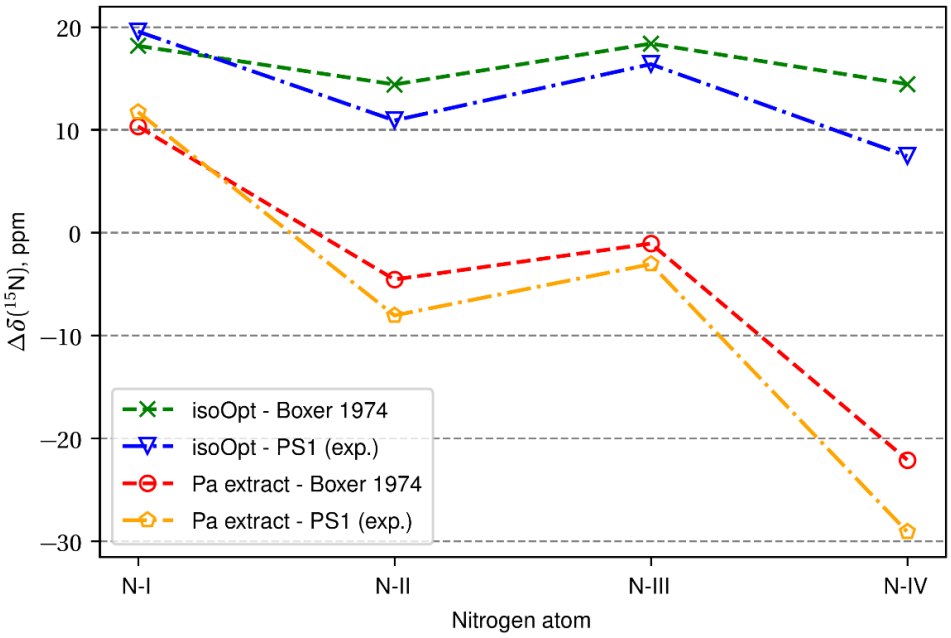

Figure S2.4: Comparison of differences in the ^{15}N Chemical shifts for "isoOpt" and "Pa extract" calculated with ADF with experimental results. The green line shows the difference of the "isoOpt" results and the results of Boxer et. al., while the blue line shows the difference towards the results of this works' authors. The orange and red line show the same differences obtained for the "Pa extract" model.

3. Chemical shifts calculated by quantum-chemical methods

Table S2.1: ^{15}N NMR shifts of P_A and P_B in ppm, including ring current effects induced by the neighboring Chl co-factor. $\Delta\delta$ is the environment effect on the shift in ppm calculated according to Eq. (1). The indices are chosen according to IUPAC (see Fig. S2.2).

index	δ(P_A), ppm			δ(P_B), ppm			Δδ, ppm	
	iso	r32	r34	iso	r32	r34	P_A	P_B
N_I	184.69	191.64	191.70	181.88	186.47	186.60	7.01	4.72
N_{II}	186.58	193.35	193.91	186.90	186.14	186.10	7.33	-0.88
N_{III}	171.88	177.28	178.38	168.13	176.57	176.52	6.50	8.39
N_{IV}	209.79	215.12	214.86	214.71	222.92	222.67	5.07	7.96

Table S2.2: ^{15}N NMR shifts of A_{0A} and A_{0B} in ppm, including ring current effects by the neighboring Chl co-factor. $\Delta\delta$ is the environment effect on the shift in ppm calculated according to Eq. (1). The indices are chosen according to IUPAC (see Fig. S2.2).

index	δ(A_{0A}), ppm			δ(A_{0B}), ppm			Δδ, ppm	
	iso	r32	r34	iso	r32	r34	A_{0A}	A_{0B}
N_I	176.12	181.65	183.29	181.38	185.08	185.28	7.17	3.90
N_{II}	185.82	182.17	182.27	186.63	194.06	194.04	-3.55	7.41
N_{III}	163.71	161.67	162.00	168.43	164.43	163.34	-1.71	-5.09
N_{IV}	204.25	206.59	206.43	209.16	208.23	208.81	2.18	-0.35

Table S2.3: ^{13}C NMR shifts of P_A and P_B in ppm, including ring current effects induced by the neighboring Chl co-factor. $\Delta\delta$ is the environment effect on the shift in ppm calculated according to Eq. (1). The indices are chosen according to IUPAC (see Fig. S2.2).

index	δ(P_A), ppm			δ(P_B), ppm			Δδ, ppm	
	iso	r32	r34	iso	r32	r34	P_A	P_B
1	138.95	140.05	140.50	142.10	142.83	142.74	1.55	0.64
2	118.54	120.73	119.14	121.18	123.20	122.92	0.60	1.74
2^1	11.91	12.65	11.01	12.03	12.97	12.96	-0.90	0.93
3	132.04	132.43	134.87	135.29	136.33	136.22	2.83	0.93
3^1	134.99	137.82	138.09	135.07	138.05	138.13	3.10	3.06
3^2	108.91	113.28	118.52	112.90	116.33	116.15	9.61	3.25
4	146.78	149.05	149.07	146.25	146.67	146.61	2.29	0.36
5	126.57	130.70	131.19	125.22	123.22	123.01	4.62	-2.21
6	150.56	154.64	154.76	151.27	154.01	153.86	4.20	2.59
7	125.93	128.86	128.94	128.31	131.85	131.72	3.01	3.41
7^1	13.78	14.49	14.43	14.24	16.31	16.30	0.65	2.06
8	139.27	140.66	141.25	138.89	140.72	140.90	1.98	2.01
8^1	27.43	27.50	27.62	27.65	27.87	27.82	0.19	0.17
8^2	29.31	30.52	31.01	23.94	24.43	24.40	1.70	0.46
9	146.12	147.28	147.40	147.92	149.10	149.31	1.28	1.39
10	122.08	122.47	122.97	118.50	118.02	119.21	0.89	0.71
11	145.38	146.03	146.27	149.08	151.09	151.07	0.89	1.99
12	128.57	130.80	132.63	128.20	136.61	137.32	4.06	9.12
12^1	16.56	14.26	14.62	16.28	16.38	16.56	-1.94	0.28
13	158.61	158.27	159.17	152.97	155.04	155.12	0.56	2.15
13^1	248.75	249.15	247.63	254.46	255.10	254.89	-1.12	0.43
13^2	65.44	66.25	66.17	80.27	79.94	79.87	0.73	-0.40
13^3	181.20	179.07	181.72	180.95	183.55	183.56	0.52	2.61
13^4	55.50	58.28	57.53	56.04	57.58	57.54	2.03	1.50
14	134.68	135.14	135.42	134.76	135.50	135.49	0.74	0.73
15	115.62	115.84	114.38	115.93	115.61	115.40	-1.24	-0.53
16	152.40	151.13	150.81	154.60	152.19	151.87	-1.59	-2.73
17	52.74	52.20	53.25	58.12	58.36	58.32	0.51	0.20
17^1	40.28	39.19	39.60	40.62	40.07	39.99	-0.68	-0.63
17^2	27.61	27.99	27.80	40.48	40.77	40.76	0.19	0.28
17^3	180.94	183.84	184.48	180.15	182.89	183.27	3.54	3.12
18	52.72	52.36	52.50	56.34	56.22	56.25	-0.22	-0.09
18^1	35.85	34.72	35.39	33.62	33.01	33.03	-0.46	-0.59
19	151.77	150.71	151.54	149.27	146.92	146.71	-0.23	-2.56
20	97.65	95.27	97.33	99.04	99.33	99.05	-0.32	0.01

Table S2.4: ^{13}C NMR shifts of A_{0A} and A_{0B} in ppm, including ring current effects induced by the neighboring Chl co-factor. $\Delta\delta$ is the environment effect on the shift in ppm calculated according to Eq. (1). The indices are chosen according to IUPAC (see Fig S2.2).

index	$\delta(A_{0A})$, ppm			$\delta(A_{0B})$, ppm			$\Delta\delta$, ppm	
	iso	r32	r34	iso	r32	r34	A_{0A}	A_{0B}
1	140.07	141.91	141.12	145.16	147.14	147.40	1.05	2.24
2	120.74	128.20	126.83	122.73	125.23	125.20	6.09	2.47
2^1	17.79	20.40	20.91	23.28	24.39	24.38	3.12	1.10
3	134.76	140.04	138.55	134.53	137.11	137.36	3.79	2.83
3^1	133.33	139.08	138.38	132.17	134.58	134.34	5.05	2.17
3^2	111.83	109.27	108.08	113.10	105.55	106.44	-3.75	-6.66
4	147.63	152.40	152.39	148.45	150.42	150.58	4.76	2.13
5	120.67	119.79	121.92	121.99	121.76	121.84	1.25	-0.15
6	143.59	143.36	144.32	145.21	148.05	148.00	0.73	2.79
7	126.31	124.20	127.69	129.45	135.17	135.01	1.38	5.56
7^1	16.31	16.14	15.73	18.65	15.88	15.87	-0.58	-2.78
8	133.86	134.20	134.30	131.45	132.34	132.13	0.44	0.68
8^1	28.25	28.47	28.59	31.18	31.00	31.06	0.34	-0.12
8^2	16.06	16.47	16.54	16.67	17.81	17.82	0.48	1.15
9	151.98	149.11	150.02	149.24	149.91	149.75	-1.98	0.51
10	123.45	117.26	117.92	122.52	119.43	118.69	-5.53	-3.83
11	152.65	151.87	151.94	154.83	151.17	150.85	-0.71	-3.98
12	130.53	131.48	131.47	137.49	137.19	137.40	0.94	-0.09
12^1	15.39	17.20	17.17	16.29	17.60	17.61	1.78	1.32
13	152.61	153.14	153.10	149.77	150.34	150.30	0.49	0.53
13^1	253.12	252.05	251.55	227.16	224.87	224.39	-1.57	-2.77
13^2	72.68	72.06	71.87	72.95	72.68	72.44	-0.81	-0.51
13^3	172.22	173.48	173.31	180.10	180.02	180.44	1.09	0.34
13^4	53.62	53.81	53.73	62.72	61.70	61.94	0.11	-0.78
14	133.67	133.47	133.43	134.57	134.07	133.97	-0.24	-0.60
15	120.96	120.25	119.98	114.90	114.32	113.20	-0.98	-1.70
16	156.54	157.17	156.43	156.09	156.03	156.25	-0.11	0.16
17	59.18	59.51	59.20	59.78	59.74	59.82	0.02	0.04
17^1	38.69	38.65	38.73	40.36	40.43	40.36	0.04	0.00
17^2	31.78	31.84	32.17	47.74	47.93	47.85	0.39	0.11
17^3	187.37	187.83	188.20	184.85	184.07	184.11	0.83	-0.74
18	45.98	45.83	45.89	48.66	48.01	48.03	-0.09	-0.63
18^1	37.81	37.97	37.80	38.05	38.25	38.24	-0.01	0.19
19	147.37	148.34	147.45	155.57	154.99	155.27	0.08	-0.30
20	102.19	103.22	102.57	101.67	102.96	103.16	0.38	1.49

4. Effect of the Protein Environment on the calculated NMR shifts

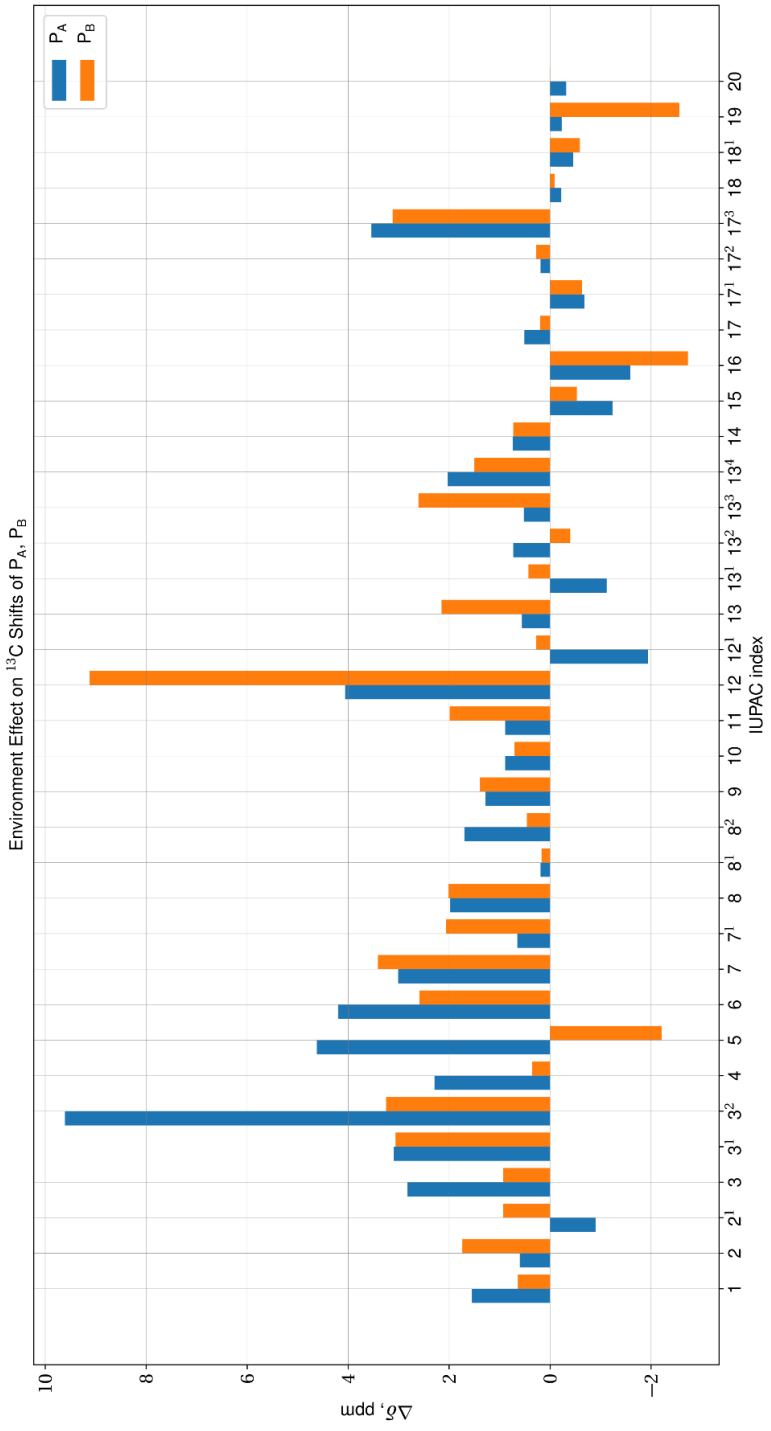

Figure S3.1: Environment effect on ^{13}C shifts of P_A and P_B.

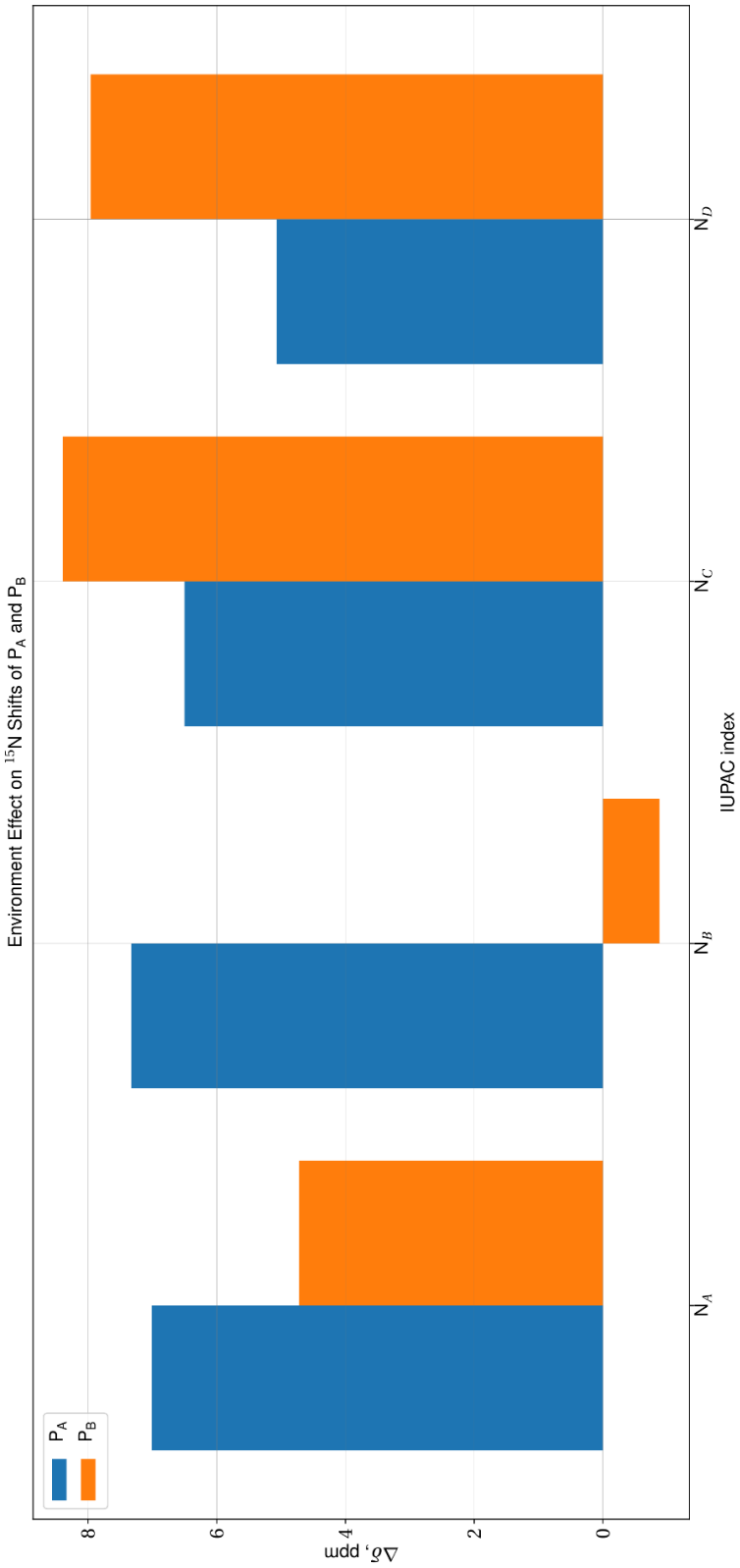

Figure S3.2: Environment effect on ^{15}N shifts of P_A and P_B.

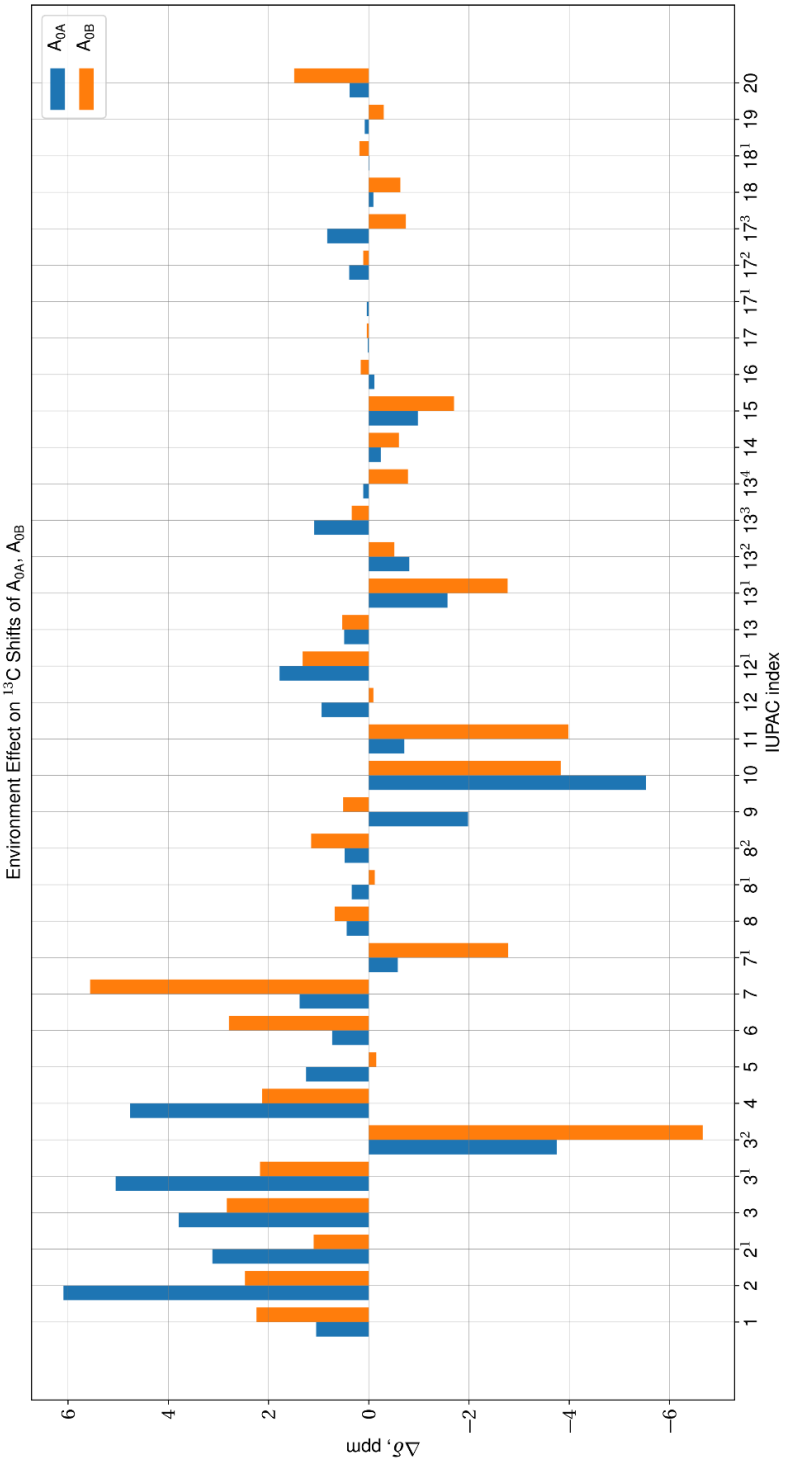

Figure S3.3: Environment effect on ^{13}C shifts of A_{0A} and A_{0B}.

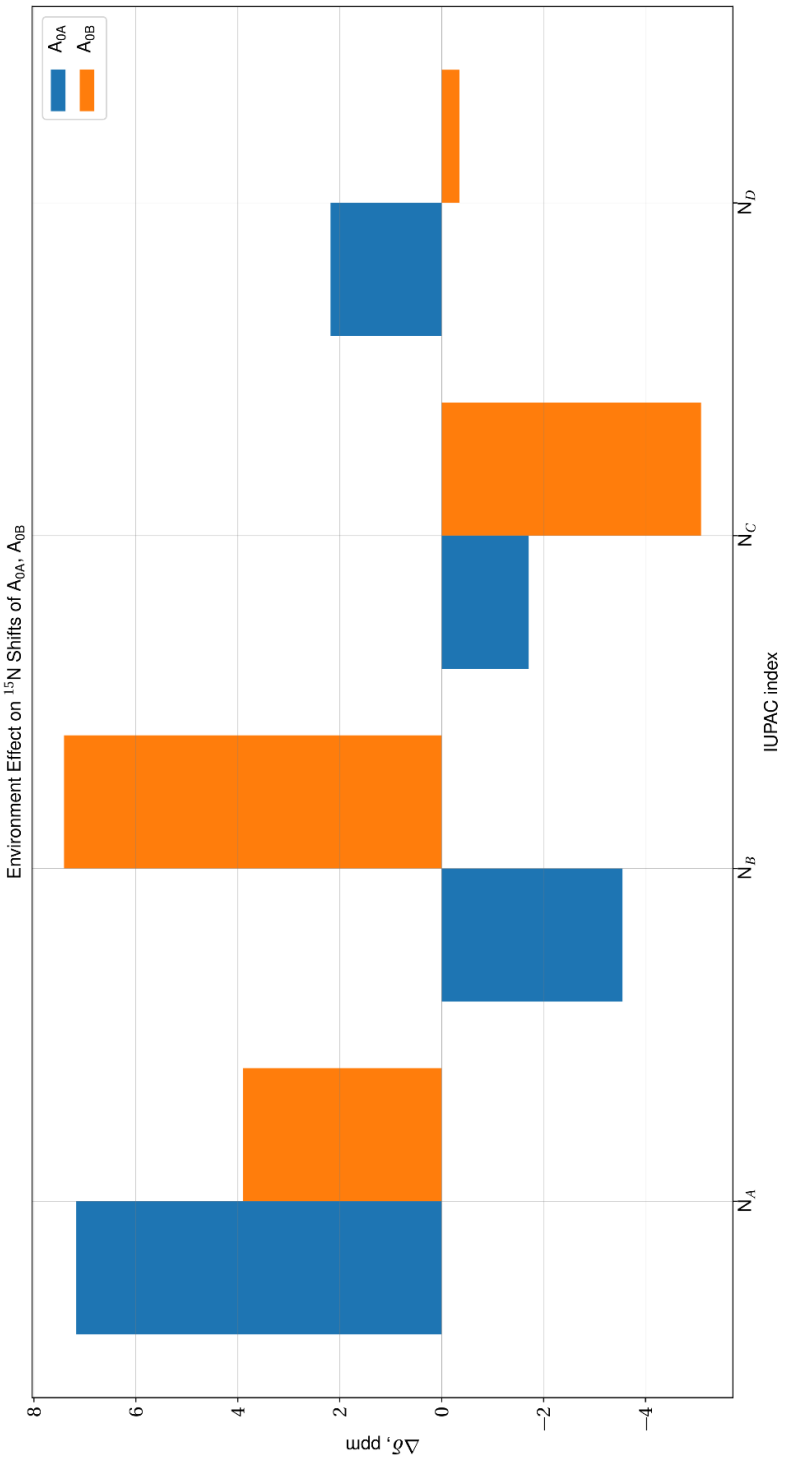

Figure S3.4: Environment effect on ^{15}N shifts of A_{0A} and A_{0B}.

5 Graphical Examples of Structures

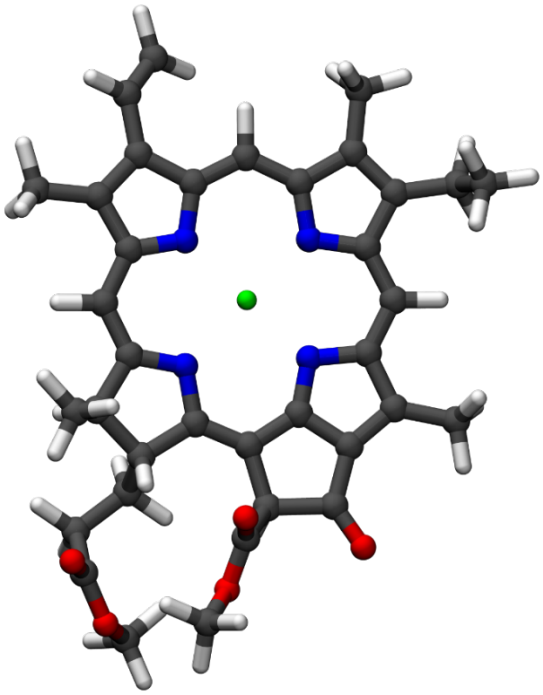

Figure S4.1: Top view of truncated Chl *a*, "isoOpt".

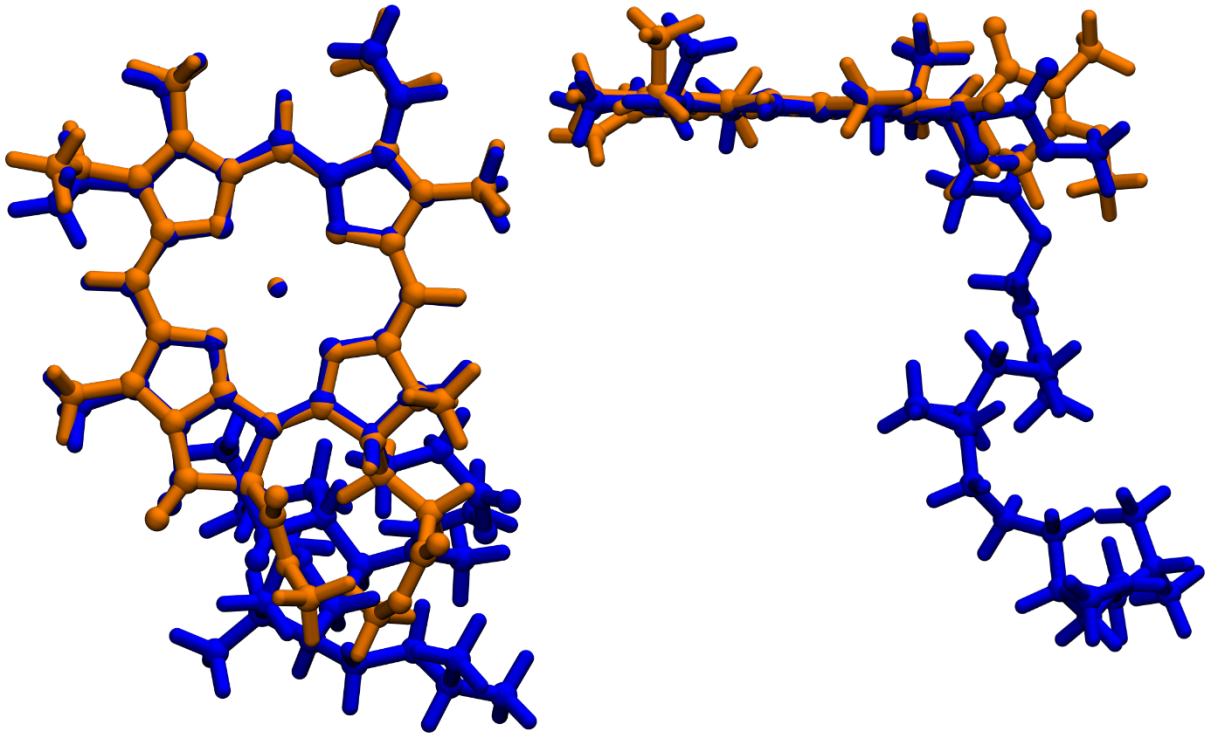

Figure S4.2: Structural comparison of the fully optimized, truncated Chl *a* "isoOpt" (orange) and "P_A extract" (blue).

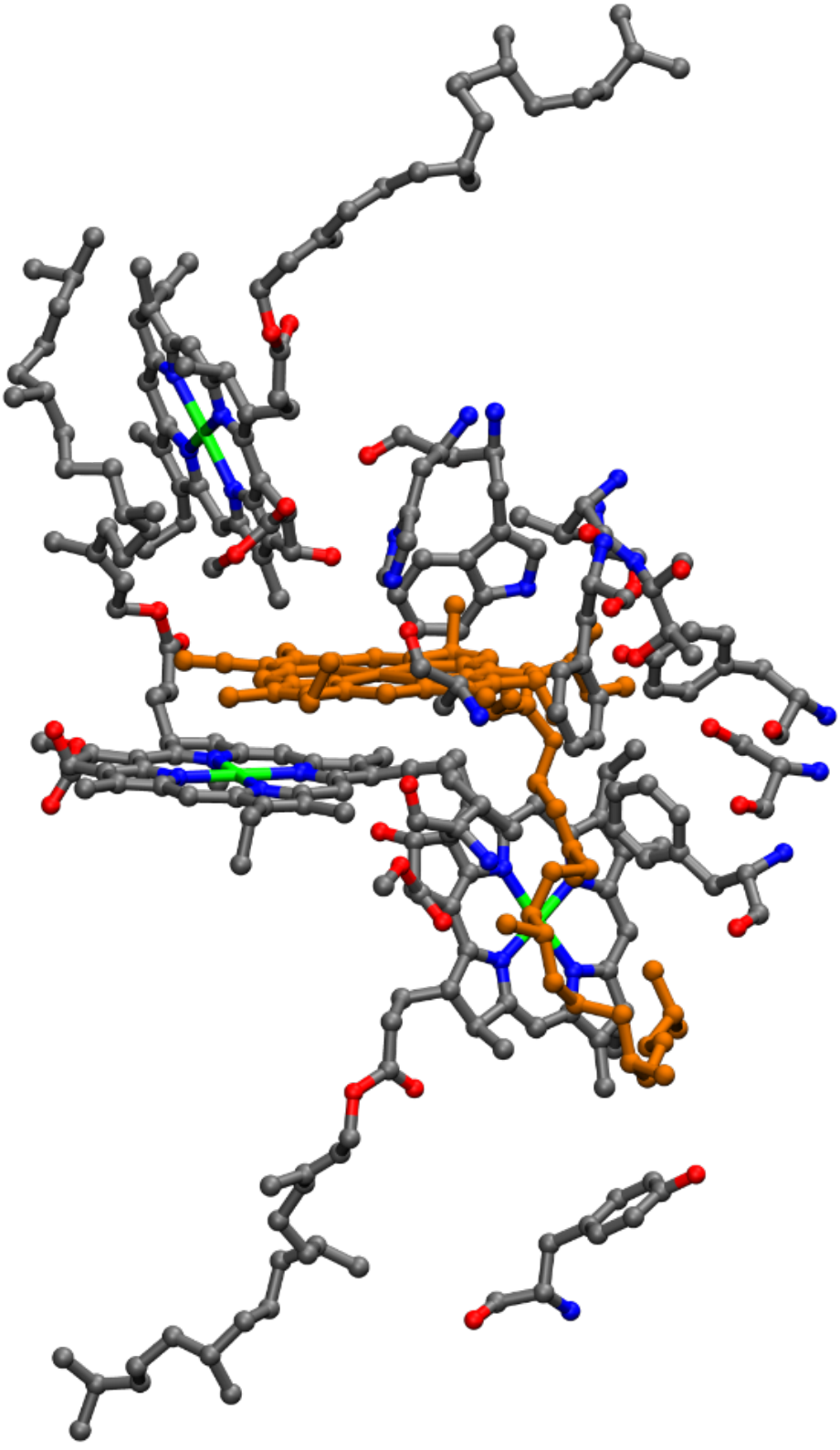

Figure S4.3: Graphical representation of the model P_A r34. The co-factor P_A is shown in orange, while the protein environment is shown in an element-wise color code. Hydrogen atoms are not shown for the sake of clarity.

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