



1 **Anatomy of unfolding: The site-specific fold stability of Yfh1 measured by 2D NMR**

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17 **Abstract**

18 Most techniques allow detection of protein unfolding either by following the behaviour of
19 single reporters or as an averaged all-or-none process. We recently added 2D NMR
20 spectroscopy to the well-established techniques able to obtain information on the process of
21 unfolding using resonances of residues in the hydrophobic core of a protein. Here, we
22 questioned whether a detailed analysis of the individual stability curves from each resonance
23 could provide additional site-specific information. We used the Yfh1 protein that has the
24 unique feature to undergo both cold and heat denaturation at temperatures above water freezing
25 at low ionic strength. We show that stability curves inconsistent with the average NMR curve
26 from hydrophobic core residues mainly comprise exposed outliers that do nevertheless provide
27 precious information. By monitoring both cold and heat denaturation of individual residues we
28 gain knowledge on the process of cold denaturation and convincingly demonstrate that the two
29 unfolding processes are intrinsically different.

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31



32 **Introduction**

33 Most techniques employed to monitor protein stability are not “regiospecific”, as they yield a
34 global result, i.e. an estimate of the stability of the whole protein architecture, observable
35 through the global evolution of secondary structure elements upon an environmental insult.
36 This is because we postulate an all-or-none cooperative process in which the protein collapses
37 altogether from a folded to an unfolded state. When monitoring the unfolding of a protein by
38 CD spectroscopy, for instance, we observe intensity changes related to the disruption of α
39 helices and/or β sheets under the influence of physical or chemical agents.

40 It is nevertheless interesting to gauge the response of selected regions of the protein or
41 even single residues during the unfolding process to gain new insights into the mechanisms of
42 unfolding in selected parts of the protein structure. A technique ideally suited for this purpose
43 is 2D NMR spectroscopy since it permits to monitor changes in the resonances at the level of
44 individual residues. Particularly suitable are 2D ^{15}N HSQC spectra since they provide a direct
45 fingerprint of the protein through mapping each of the amide protons. Volume variations of the
46 NMR resonances may reflect changes affecting single atoms of each residue and indirectly
47 report on how they are individually affected by the unfolding process. We have recently shown
48 that it is possible to use 2D NMR to measure protein stability and get thermodynamics
49 parameters comparable to those obtained by standard CD methods, provided that a suitable
50 selection of the residues is made, followed by the subsequent average of the changes of these
51 residues (Puglisi et al., 2020). To choose residues whose NH is deep inside the protein and
52 relatively inaccessible to the solvent we employed SADIC (Varrazzo et al., 2005), a software
53 that quantifies depth inside the protein (D), in combination with PopS (Cavallo et al, 2003) that
54 yields relative accessibility at an atomic level (RA). We combined the two parameters defining
55 a new parameter, RAD, which combines depth and exposure. We demonstrated that the
56 stability curve calculated from averaging amide volumes from residues with a RAD value
57 below 0.1 (here henceforth called RAD_0.1) is consistent with that calculated from CD
58 spectroscopy (Puglisi et al., 2020).

59 It is now interesting to wonder what information is carried by residues far from the
60 hydrophobic core and how they reflect the process of unfolding. This is relevant also in view
61 of an increasing number of studies on protein stability based on the intensity variations of the
62 resonance of a single residue upon unfolding (Danielsson et al., 2015; Smith et al., 20116;
63 Guseman et al., 2018). The excellent agreement between NMR and CD thermodynamic
64 parameters (Puglisi et al., 2020) put us in the position to examine the output of single residues
65 critically, and eventually follow the process of unfolding at an atomic level.



66 Here, we present the analysis of the stability of the yeast ortholog, Yfh1, of human frataxin as
67 measured by the stability curves of most observable, isolated NH resonances. We chose Yfh1
68 to probe regiospecific unfolding because this protein is an ideal model system for measuring
69 stability curves of single residues: in addition to heat denaturation, Yfh1 has a cold denaturation
70 temperature observable above zero degrees when in the absence of salt (Pastore et al., 2007).
71 Observation of the two unfolding temperatures facilitates enormously the calculation of reliable
72 stability curves and of the whole set of thermodynamic parameters. The usefulness of Yfh1 as
73 a tool to investigate unfolding processes is evidenced not only by our subsequent work (Pastore
74 et al., 2007; Sanfelice et al., 2013; Pastore and Temussi, 2017; Martin et al., 2008; Sanfelice et
75 al., 2014; Alfano et al., 2017) but also by papers from other laboratories (Espinosa et al., 2016;
76 Chatterjee et al., 2014; Bonetti et al., 2014; Aznauryan et al., 2013).

77 We demonstrate that it is possible to sort out which individual single residues yield
78 stability curves consistent with the global unfolding process and that we can obtain valuable
79 information on the process of unfolding from residues that diverge from the average behaviour,
80 Our data also prove directly the distinct mechanisms determining the cold and heat denaturation
81 processes by providing site-specific information on solvent interactions.

82

83 **Results**

84 We collected ^{15}N HSQC spectra of Yfh1 at different temperatures and from them plotted the
85 volumes of individual residues as a function of temperature. It is possible to extract
86 thermodynamic parameters from these plots provided some conditions are met (Privalov, 1990;
87 Martin et al., 2008). It is first assumed that unfolding transitions are two-state processes from
88 folded (F) to unfolded (U) states. It is then hypothesized that the difference of the heat capacity
89 of the two forms (ΔC_p) does not depend on temperature. This assumption is considered
90 reasonable when the heat capacities of the native and denatured states change in parallel with
91 temperature variation (Privalov, 1990). When these conditions are true, the populations of the
92 two states at temperature T, $f_F(T)$ and $f_U(T)$, can be expressed as a function of the difference in
93 free energy, $\Delta G^\circ(T)$, according to the modified Gibbs-Helmholtz equation (Martin et al., 2008).
94 The curve corresponding to this equation is known as the stability curve of the protein (Becktel
95 and Schellman, 1987). The main thermodynamic parameters, T_m , ΔH_m and ΔC_p , can be
96 determined using a non-linear fit (damped least-squares method). Other parameters for low
97 temperature unfolding, e.g. T_c , can be read from the stability curve. Volumes of isolated
98 residues were transformed into relative populations of folded Yfh1 assuming that, as found in



99 other studies on Yfh1 (Pastore et al., 2007; Martin et al., 2008; Sanfelice et al., 2014; Alfano
100 et al., 2017), unfolded forms are in equilibrium with a 70% population of folded Yfh1 at room
101 temperature (**Table 1**). The concurrent presence of an equilibrium between folded and
102 unfolding species is directly testified by the spectra at low ionic strength: folded and well
103 dispersed resonances co-exist with others strongly overlapping. These extra peaks disappear as
104 soon as physiologic concentrations of NaCl are added (Vilanova et al., 2014).

105 We then correlated each amide resonance to the corresponding value of RAD, the
106 parameter introduced in Puglisi et al. (2020), to pinpoint residues close to the hydrophobic
107 core. The behaviour of resonances in the HSQC spectrum of Yfh1 as a function of temperature
108 was also not uniform. While some peaks could be observed nearly at all temperatures in the
109 range 273-323 K, others disappeared at temperatures intermediate between room temperature
110 and the two unfolding temperatures, i.e. lower than 323 K or higher than 273 K (**Figure S1 of**
111 **Suppl. Mat.**). This behaviour was of course related to the exchange regime of these residues
112 and told us that they are not an integral part of the architecture of the folded form and thus their
113 volume variations cannot represent the all-or-none overall unfolding process faithfully. It was
114 anyway possible to calculate a stability curve from the temperature dependence of the
115 resonance volumes and the corresponding thermodynamic parameters for many residues, to
116 yield valuable information on the unfolding process. We then looked into what these residues
117 could tell us about the unfolding process.

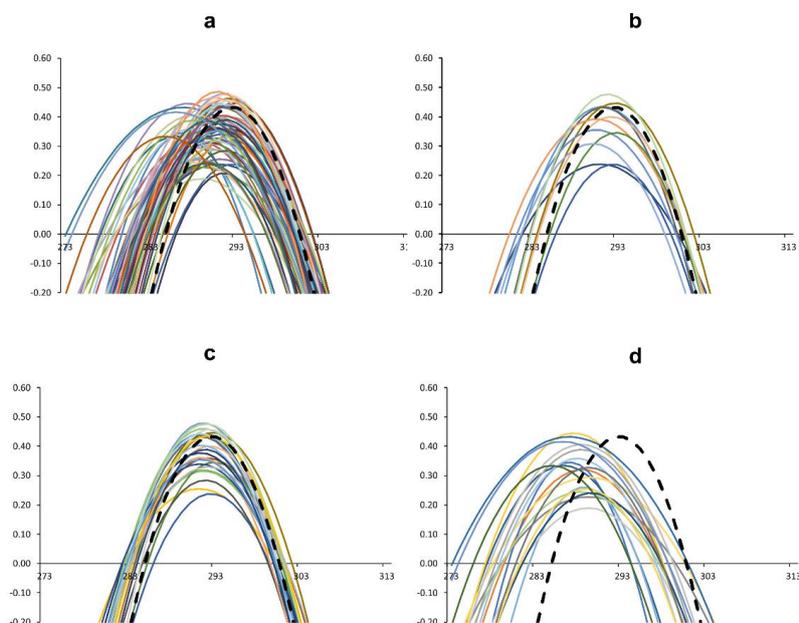
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119 **Residues consistent with or outliers from the global behaviour**

120 Comparison of the stability curves of all the well behaved residues (68 over the expected 109
121 resonances) with the average best curves calculated for RAD values <0.1 (henceforth called
122 RAD_{0.1}) showed that several residues yield stability curves drastically different from the
123 average curve (**Figures 1a**). The curves for residues in the hydrophobic core are overall in good
124 agreement with the best average curve (**Figures 1b**). However, there is in principle no clear-
125 cut criterion to decide when the curves are not consistent with the average. We arbitrarily chose
126 to set a cut-off at values of the unfolding temperatures (T_m and T_c) that differed, on average,
127 less than 1.5 K from those corresponding to the average (RAD_{0.1}). This difference is smaller
128 than the variability that we had observed among different preparations and measurements of
129 the same protein (Pastore et al., 2007; Martin et al., 2008; Sanfelice et al., 2014; Sanfelice et
130 al., 2015; Alfano et al., 2017; Puglisi et al., 2020). The residues selected according to this
131 criterion are E71, E75, E89, L91, D101, L104, S105, M109, T110, F116, Y119, I130, L132,
132 A133, F142, D143, L152, L158, T159, D160, T163, and K168 (**Figure 1c**). Most of the amide



133 groups of the well-behaved residues are spread among well-structured secondary elements, but
134 there are a few in less ordered regions (**Figure 2a**). By the same token, we selected as ‘ill-
135 behaved’ residues those whose ΔT_m and ΔT_c were greater, on average, than 3 K with respect
136 to the best curve (RAD_0.1). Eighteen residues (V61, Q63, H83, H95, C98, G107, V108, I113,
137 V120, N127, K128, Q129, L136, N146, G147, N154, K172, Q174) belong to this sub-set.
138 Similarly, except for a few outliers, they all are in less structured regions (**Figure 2b**).
139



140

141 **Figure 1.** Comparison of single residue stability curves with the global (RAD_0.1) best curve
142 (dashed black). **a**) Stability curves of all observable isolated residues. **b**) Stability curves of
143 residues with a RAD < 0.1. **c**) Stability curves of single residues for which the difference in the
144 unfolding temperatures with respect to values of the reference curve (ΔT_m and ΔT_c) is on
145 average below 1.5 °C **d**) Stability curves of single residues for which the difference in the
146 unfolding temperatures with respect to values of the average curve (ΔT_m and ΔT_c) is on
147 average above 3 K.

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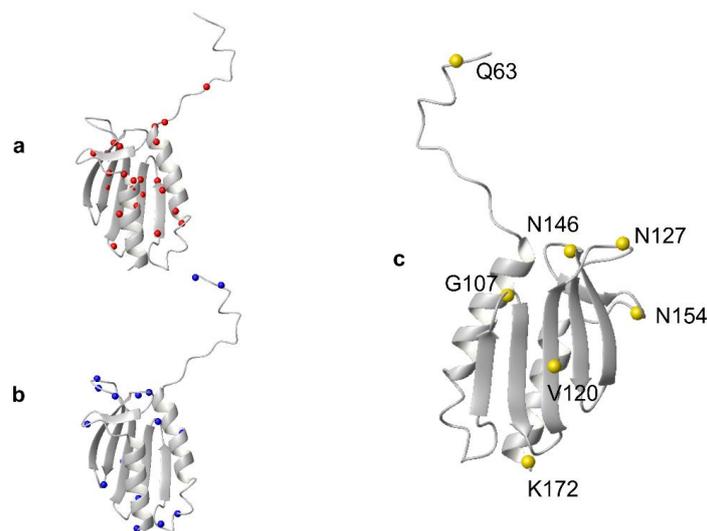
149 The stability curves of these residues (**Figure 1d**) have an important peculiarity: most stability
150 curves show a moderate decrease of T_m and a large decrease of T_c . This finding is paradoxical
151 because it implies that the corresponding transition temperatures for the heat and cold unfolding
152 point to decreased and increased stability for heat and cold denaturation respectively. Although
153 it is difficult to explain the behaviour of the extreme values of the ΔT_c of some residues, it is
154 fair to say that this behaviour confirms that the mechanisms of the two unfolding processes are
155 intrinsically different. This possibility was already postulated by Privalov (Privalov, 1990) who



156 suggested that the disruption of the hydrophobic core at low temperature would be caused by
157 the hydration of the side chains of hydrophobic residues of the core, whereas the high
158 temperature transition is mainly linked to entropic reasons, consistent with the increase of
159 thermal motions when temperature is increased. What we observed is also in line with our
160 previous evidence that showed that the unfolded species at low temperature has a volume
161 higher than the folded species and of the high temperature unfolded species (Alfano et al.,
162 2017) and that cold denaturation is caused by a hydration increase (Adrover et al., 2012).

163 In light of this fundamental difference, it is tempting to hypothesize that the large ΔT_c of
164 residues positioned in the middle of connecting turns, G107, N127, N146 and N154 (**Figure**
165 **2c**), may reflect the fact that these flexible (well hydrated) structural elements keep some
166 resilience since they do not experience environmental changes even after the core has been
167 invaded by water molecules and are the last to be affected by unfolding.

168



169

170 **Figure 2.** Distribution of residues on the structure of Yfh1. a) Distribution of the N atoms of
171 residues for which the difference in the unfolding temperatures with respect to values of the
172 RAD_0.1 curve (ΔT_m and ΔT_c) is on average below 1.5 K. b) Distribution of the N atoms of
173 residues for which the difference in the unfolding temperatures with respect to values of the
174 average curve (ΔT_m and ΔT_c) is on average above 3 K. c) The gold dots on the structure mark
175 the positions of residues whose stability curve is most shifted to lower temperatures with
176 respect to the average one (RAD_0.1).

177

178 In other words, these residues seem to form a kind of exoskeleton that is tougher at lower
179 temperatures. This view is consistent with the observation that, when decreasing the

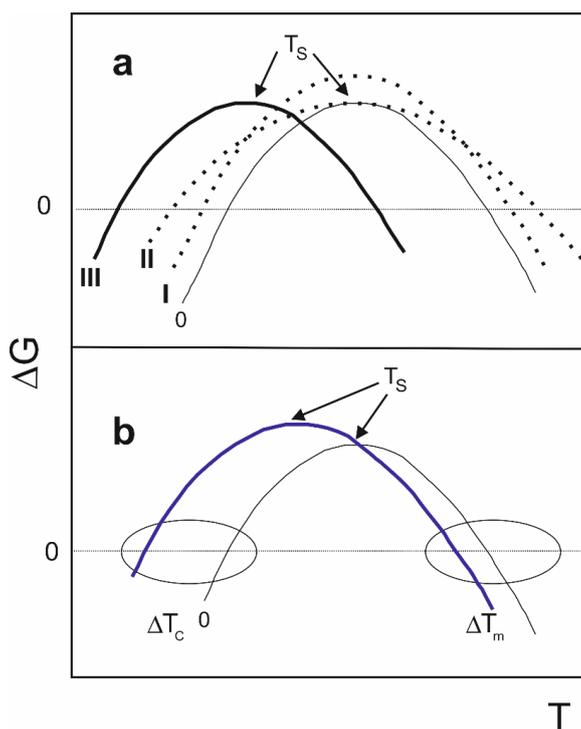


180 temperature towards the cold denaturation transition there is a decrease of thermal motions that
181 may favour the persistence of structural elements less ordered than helices and beta sheet.

182

183 **A thermodynamic assessment of flexibility**

184 The negative ΔT_m and ΔT_c observed for some residues (**Figure 1d**) imply that the temperature
185 of maximum stability (T_S , so called because it corresponds to zero entropy) is lower than that
186 observed for the best average (RAD_0.1). The low temperature shift of T_S values is consistent
187 with the unfolding of more flexible parts of the protein structure because it corresponds to an
188 increase in entropy connected to unfolding.



189

190 **Figure 3.** Mechanisms that influence stability curves of a protein (adapted from Nojima
191 et al, 1977). **a)** Dependence of the difference of free energy between unfolded and folded
192 states (ΔG) of a hypothetical protein vs temperature (curve 0). Mechanism I illustrates
193 the effect of increasing ΔH_s (curve I). Mechanism II shows the effect of reducing ΔC_p
194 (curve II). Mechanism III shows the shift of the whole stability curve towards lower
195 temperatures caused by increasing ΔS_m (curve III). **b)** A combination of the three
196 mechanisms. The solid blue curve corresponds qualitatively to the cases of Yfh1 reported
197 in **Figure 1 d**.

198



199 This consideration (Razvi & Scholtz, 2006) is based on the classification by Nojima et
200 al. (1977) of the main mechanisms of changing the thermal resistance of a protein.
201 Obviously, the same holds true if one wants to *decrease* T_m or T_c : it is sufficient to reverse the
202 changes. According to the rough classification of Nojima et al. (1977), altered thermostability
203 can be achieved thermodynamically in three different ways (**Figure 3**).
204 According to mechanism (I), when ΔH_s (the change in enthalpy measured at T_s) increases, the
205 stability curve retains the same shape, but with greater ΔG values at all temperatures. With
206 mechanism II, a decreased ΔC_p leads to a broadened stability curve, because the curvature of
207 the stability curve is given (Becktel, & Schellman, 1987) by $\frac{\partial^2 \Delta G}{\partial T^2} = -\frac{\Delta C_p}{T}$. According to
208 mechanism III, the entire curve can shift towards higher or lower temperatures. It is possible
209 to show (Privalov, 1990) that:

$$210 \quad T_s = T_m \cdot \exp\left[-\frac{\Delta S_m}{\Delta C_p}\right] = T_m \cdot \exp\left[-\frac{\Delta H_m}{T_m \cdot \Delta C_p}\right].$$

211 Increasing the difference in entropy between the folded and unfolded states (ΔS_m) can shift
212 values of T_s towards lower temperatures. Most of the curves of **Figure 1d** do not correspond
213 to a single mechanism, but to a combination of them (**Figure 3b**). However, all are shifted
214 toward lower values of T_s . The largest low-temperature differences correlate well with less
215 ordered regions of the structure. These regions experience largest unfolding entropies and thus
216 visit a larger number of conformations. It is not surprising to find this behaviour for residues
217 at the N- and C-termini (Q63 and K172) or in connecting loops (G107, N127, N146 and N154)
218 which are bound to be flexible (Halle, 2002). More surprising is, however, to find amongst
219 these residues also V120 which is right in the middle of the β -sheet. While we have not a
220 definite explanation for this observation at the moment, it could indicate a local frustration
221 point in this region.

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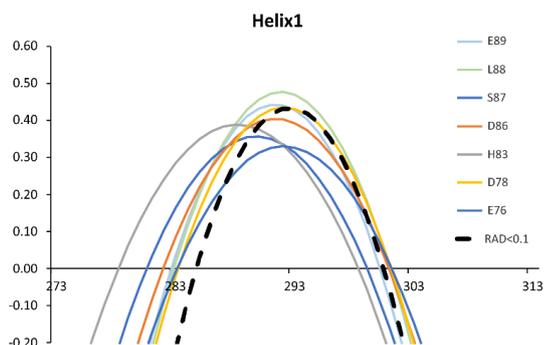
223 **Exploring the correlation between stability and secondary structure elements**

224 We have previously shown that, in addition to the criteria of depth and exposition, an
225 alternative selection of residues over which average populations might be based on elements
226 of regular secondary structure (Puglisi et al., 2020). It is now possible to analyse the behaviour
227 of each secondary structure element. The stability curves related to accessible residues of
228 individual secondary structure elements are summarized in **Figures 4** and **5**. The largest

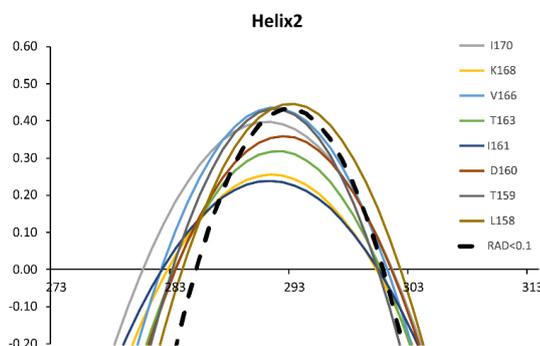


229 number of residues of secondary structure traits whose resonance is accessible belongs to the
230 two helices (**Figure 4**).

a



b



231

232 **Figure 4.** Stability curves of residues belonging to secondary structure elements. **a)** Helix 1.
233 **b)** Helix 2. Residues are labelled with single letter code. The average stability curve is shown
234 as black dashed line.

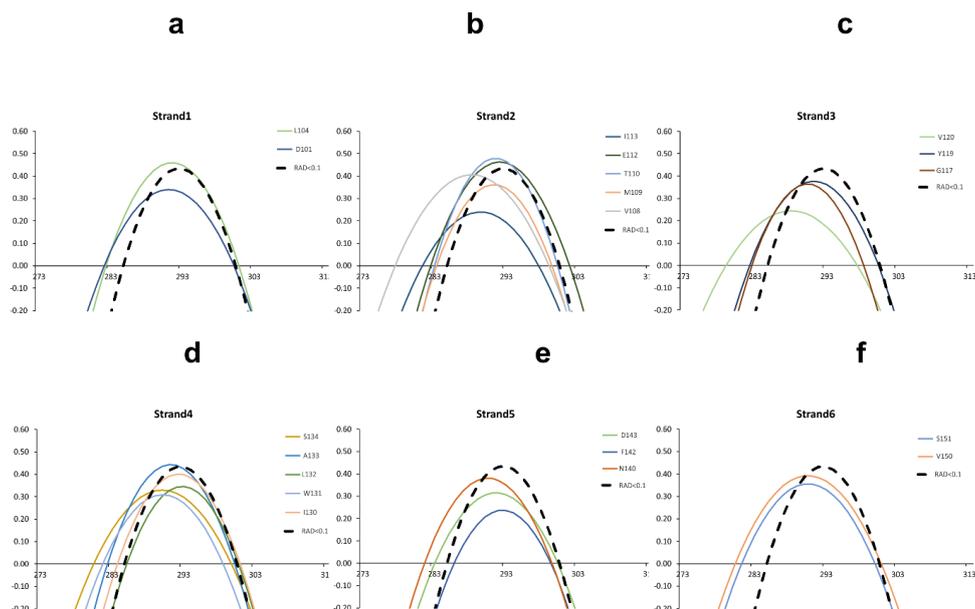
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236 Both for helix 1 (**a**) and helix 2 (**b**) there are several resonances whose stability curve is far
237 from the reference one (dashed black curve of RAD_0.1). In particular, these are those of His83
238 and Ser87 for helix 1. All the others are in fair agreement with the average curve. The best
239 behaved residues (Glu76, Asp78, Leu88 and Glu89) are located at the two ends of the helix
240 with their amide groups in the buried side of the helix. For helix 2, the worst agreement is found
241 for Ile161 and Ile170, whereas the best agreement is for Leu158, Thr159, Asp160, Thr163 and
242 Lys168. This implies that residues of helix 2 with a good agreement are distributed over the
243 whole secondary structure element.

244 The number of residues belonging to beta strands for which it was possible to extract
245 stability curves is more limited (**Figure 5**).



246



247

Figure 5. Stability curves of residues belonging to secondary structure elements. a) Strand 1. b) Strand 2. c) Strand 3. d) Strand 4. e) Strand 5. f) Strand 6. Residues are labelled with single letter code. The average stability curve is shown as black dashed line.

251

The best agreement was found for Leu104 of strand 1, Met109 and Thr110 of strand 2, Ile130, Leu132 and Ala133 of strand 4 and Phe142 and Asp143 of strand 5. It is interesting to note that some of the best residues reported in **Figure 1c** are not present in regular secondary structure elements.

256

257 **The behaviour of tryptophan side chains**

258 We then looked into the possibility of following the process of unfolding and calculating
259 thermodynamic parameters using the tryptophan side chains. This choice directly parallels
260 studies based on following the process of unfolding by fluorescence using the intrinsic
261 tryptophan fluorescence (Monsellier & Bedouelle, 2002). Yfh1 has two tryptophans: W131 is
262 fully exposed to the solvent whereas W149 is buried. Both residues are fully conserved
263 throughout the frataxin family and the two side chain resonances are clearly identifiable
264 (**Figure S2a of Suppl. Mat.**). We calculated the thermodynamic parameters for the side chain
265 indole groups of both residues (**Table 1**) by the same procedure outlined for main chain NHs,
266 generating first a stability curve (**Figure S2b of Suppl. Mat.**). The resonance of W149, which
267 could potentially be more interesting, could not be used for quantitative measurements because



268 the temperature dependence of its volume yields a stability curve very different from the others
269 (**Figure S2b of Suppl. Mat.**) which leads to impossible parameters. This might be explained
270 by the co-existence of folded and partially unfolded species in equilibrium with each other in
271 solution. As a consequence the indole of W149 resonates both at 9.25 and 127.00 ppm (folded
272 specie) and at ca. 10.05 and 129.20 ppm (split into three closely adjacent peaks, unfolding
273 intermediates) (**Figure S2a of Suppl. Mat.**). As previously proven experimentally, the
274 resonances of the unfolding intermediates disappear upon addition of salt (Figure 1, panel A
275 and B in Vilanova et al., 2014). These resonances are also at the same coordinates observed for
276 the tryptophan indole groups at low and high temperature where however the three species
277 collapse into one (**Figure S1 of Suppl. Mat.**). The complex equilibrium between different
278 species could thus explain the ill-behaviour of the corresponding stability curve of this residue.
279 The behaviour of the resonance of the exposed W131 side chain is instead fully consistent with
280 that of RAD_0.1 and also with the original curve calculated from 1D NMR data (Pastore et al.,
281 2007). On the whole, these results exemplify well the complexity of the selection choice of the
282 unfolding reporter and advocate in favour of a wholistic analysis of the whole set of available
283 data.

284

285 **Discussion**

286 The *de facto* demonstration that it is possible to reliably measure the thermodynamic
287 parameters of protein unfolding by 2D NMR spectroscopy (Puglisi et al., 2020) has opened a
288 new territory to study protein unfolding at atomic resolution using site-specific information.
289 Following protein folding/unfolding looking at specific residues rather than obtaining an
290 average overall picture is not a novelty. Despite some intrinsic limitations, fluorescence has,
291 for instance, been used for decades to probe protein unfolding following the intrinsic
292 tryptophan fluorescence (Monsellier & Bedouelle, 2002; Bolis et al., 2004). Another elegant,
293 although sadly still underexploited technique able to report local behaviour at the level of
294 specific residues is chemically induced dynamic nuclear polarization (CIDNP) first introduced
295 to the study of proteins by Robert Kaptein (Kaptein et al., 1978). This technique allows the
296 selective observation of exposed tryptophans, histidines and tyrosines. In protein folding, it
297 was, for instance, used to characterize the unfolded states of lysozyme (Broadhurst et al., 1991;
298 Schlörb et al., 2006) and the molten globule folding intermediate of α -lactalbumin (Improta et
299 al., 1995; Lyon et al., 2002). Real-time CIDNP was also used to study the refolding of
300 ribonuclease A (Day et al., 2009) and HPr (Canet et al., 2003). The only drawback of this
301 technique is that, as in fluorescence, the information is limited to specific aromatic residues.



302 Another important technique that reports on protein unfolding at the single residue level
303 is stopped-flow methods coupled with NMR measurements of hydrogen exchange (Kim and
304 Baldwin, 1991; Roder and Wüthrich, 1986) and by mass spectrometry (Miranker et al., 1993).
305 In a classic paper (Miranker et al., 1991), Dobson and co-workers described, for instance, NMR
306 experiments based on competition between hydrogen exchange as observed in COSY spectra
307 and the refolding process. The authors could conclude in this way that the two structural
308 domains of lysozyme followed two distinct folding pathways, which significantly differed in
309 the extent of compactness in the early stages of folding. Similar and complementary
310 conclusions could be reached by integrating NMR with mass spectrometry (Miranker et al.,
311 1993). While these studies retain their solid importance, the possibility of following the
312 resonance intensities also by HSQC spectra may provide a more flexible tool to obtain detailed
313 information on unfolding as it reports on the exchange regime but also, implicitly, on the
314 chemical environment. The use of 2D HSQC had been discouraged by the non-linear
315 relationship between peak intensity (or volume) and the populations with temperature as the
316 consequence of relaxation, imperfect pulses, and mismatch of the INEPT delay with specific
317 J-couplings. We have previously suggested an approach to compensate for these effects and
318 demonstrated that the non-linearity does not affect the spectra of Yfh1 (Puglisi et al., 2020),
319 even though these conclusions might be protein dependent.

320 Here, we reconsidered our previous work (Puglisi et al., 2020) and measured individual
321 stability curves for most of the residues of Yfh1. Our approach showed to be particularly
322 fruitful for the study of this protein that has an unusual if not unique behaviour since, as a
323 natural unmodified full-length protein, it undergoes cold and heat denaturation when in the
324 absence of salts, allowing measurement of the whole stability curve. The availability of this
325 model system permitted us to shed light onto several important aspects.

326 We observed that the behaviour of the individual stability curves is not distributed
327 uniformly along the sequence. Residues can be clearly divided into two groups, i.e. those
328 consistent with the average behaviour of an all-or-none mechanism of unfolding and those
329 differing, even strongly, from the best average (RAD_0.1). This finding alone proved that it is
330 not possible to measure stability using a single residue without a careful evaluation of the role
331 of the specific residue in the protein fold. This conclusion is partially mitigated by our results
332 on the parameters obtained for a tryptophan indole. However, in the whole, also for these side
333 chains it may be difficult, *a priori*, to infer which tryptophan is more reliable, thus suggesting
334 that unfolding studies based on fluorescent measurements using the intrinsic fluorescence of
335 tryptophan should be taken with a pinch of salt. Even though our findings support the



336 possibility of obtaining protein stability parameters using the intrinsic tryptophan fluorescence
337 data in our specific example, in many other cases no independent controls could be done to
338 evaluate the accuracy of the results. The possibility of using 2D NMR and the introduction of
339 the easily approachable RAD parameter may assist in this choice in future studies.

340 Analysis of individual secondary structure elements, i.e. helices and strands, showed that
341 there is no clear hierarchy among them, and there is no indication that any of the elements
342 undergoes disruption before the others, either at high or at low temperature. It will be
343 interesting in the future to study lysozyme to have an example in which two subdomains unfold
344 independently (Miranker et al., 1991). In addition to information on regular secondary structure
345 elements, our analysis yielded also interesting information on less ordered traits. Intrinsically
346 flexible elements, i.e. regions characterized by multiple conformers, can be identified
347 unequivocally by their thermodynamic parameters, without recurring to interpretative
348 mechanisms.

349 Another important point is that we observed a clear difference between parameters
350 corresponding to the cold and the heat denaturation processes: residues that are outliers from
351 the average stability curve tend to have a strong stabilization effect at low temperature and a
352 weak destabilising effect at high temperature. This is a strong confirmation that the
353 mechanisms of the two transitions are intrinsically different according to the mechanism of
354 cold unfolding proposed by Privalov. In this model, cold denaturation is intimately linked to
355 the hydration of hydrophobic residues of the core (Privalov, 1990; Adrover et al., 2012). One
356 outstanding consequence is that, at the temperature of global unfolding, corresponding to that
357 of the average of the deeply buried protein core (RAD_0.1), residues outside the hydrophobic
358 core and in regions classified as flexible may be more resilient against unfolding. In other
359 words, at low temperature, opening of the hydrophobic core and its disruption can happen
360 before the collapse of external and more exposed elements.

361 In conclusion, we can state that monitoring protein degradation by individual residue
362 stability curves, as allowed by 2D NMR spectroscopy, yields a much more informative picture
363 than what may be obtained by traditional methods, particularly when both cold and heat
364 unfolding can be observed.

365

366 **Methods**

367 *Sample preparation*



368 Yeast frataxin (Yfh1) was expressed in BL21(DE3) *E. coli* as previously described (Pastore et
369 al., 2007). To obtain uniformly ^{15}N -enriched Yfh1, bacteria were grown in M9 using ^{15}N -
370 ammonium sulphate as the only source of nitrogen until an OD of 0.6-0.8 was reached and
371 induced for 4 hours at 310 K with 0.5 mM IPTG. Purification required two precipitation steps
372 with ammonium sulphate and dialysis followed by anion exchange chromatography using a Q-
373 sepharose column with a NaCl gradient. After dialysis the protein was further purified by a
374 chromatography using a Phenyl Sepharose column with a decreasing gradient of ammonium
375 sulphate.

376

377 *NMR measurements*

378 2D NMR ^{15}N -HSQC experiments were run on a 700 MHz Bruker AVANCE spectrometer.
379 Following the strategy previously described (Puglisi et al., 2020), ^{15}N -labelled Yfh1 was
380 dissolved in 10 mM Hepes at pH 7.5 to reach 0.1 mM. Spectra were recorded in the range 5-
381 40 °C with intervals of 2.5 °C and using the Watergate water suppression sequence (Piotto et
382 al., 1992). For each increment 8 scans were accumulated, for a total of 240 increments. Spectra
383 were processed with NMRPipe and analysed with CCPNMR software. Gaussian (LB -15 and
384 GB 0.1) and cosine window functions were applied for the direct and indirect dimension
385 respectively. The data were zero-filled twice in both dimensions. Spectral assignments of Yfh1
386 correspond to the BMRB deposition entry 19991.

387

388 **Selection of the best set of amides**

389 Yfh1 contains 114 backbone amide protons. The first 23 residues are an intrinsically sequence
390 that contains the region for mitochondrial import and processing, leading to 91 resonances in
391 the globular domain. 68 have non-overlapping and isolated resonances that allow easily
392 detectable and reliable volume calculation. Most of the excluded overlapping resonances
393 corresponded to disordered regions or to a partially unfolded conformation in equilibrium with
394 the folded one in a slow exchange regime at room temperature (Sanfelice et al., 2014).

395 Volumes were calculated by summation of intensities in a set box using the CCPNMR software.
396 The parameter RAD was used taking the parameters from the software Pops
397 (<http://mathbio.nimr.mrc.ac.uk/~ffranca/POPS>) and SADIC
398 (<http://www.sbl.unisi.it/prococoa/>).

399 Residues involved in secondary structures were evaluated according to the DSSP program
400 (<https://swift.cmbi.umcn.nl/gv/dssp/>). This software is all freely available. Our analysis
401 resulted in 35 residues in secondary structure elements (15 in alpha helices, 20 in beta sheets),



402 39 residues having RAD <0.5, 37 with RAD <0.4, 33 with RAD <0.3, 24 with RAD <0.2 and
403 11 having RAD < 0.1 (Puglisi et al., 2020).

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- 526



527 **Table 1.** Thermodynamic parameters of all residues

	ΔH (Kcal/mol)	ΔS (Kcal/mol)	ΔC_p (Kcal/molK)	T_m (K)	T_c (K)	RAD
61 Val	19.94	0.067	1.58	298.41	273.82	48.20
63 Gln	21.07	0.072	2.24	294.03	275.56	3.64
64 Glu	27.70	0.093	3.30	297.96	281.43	2.52
65 Val	24.30	0.081	2.71	299.73	282.10	0.31
68 Leu	21.01	0.071	3.02	296.52	282.77	0.75
70 Leu	28.54	0.096	3.73	298.15	283.06	2.35
71 Glu	29.09	0.097	3.59	299.79	283.82	7.13
72 Lys	33.11	0.111	3.73	299.59	282.13	6.24
75 Glu	24.30	0.081	2.75	300.01	282.63	2.07
76 Glu	21.81	0.072	2.37	301.19	283.11	0.91
78 Asp	29.03	0.096	3.18	301.13	283.19	0.17
83 His	22.77	0.076	2.21	298.36	278.17	0.16
86 Asp	25.39	0.084	2.62	300.96	281.94	0.27
87 Ser	22.84	0.076	2.42	299.10	280.57	0.34
88 Leu	31.31	0.104	3.38	301.03	282.83	0.04
89 Glu	30.04	0.100	3.36	300.20	282.62	0.20
90 Glu	26.07	0.087	2.50	300.86	280.43	0.52
91 Leu	34.10	0.114	4.19	300.30	284.26	0.16
92 Ser	28.77	0.096	2.93	300.13	280.87	0.15
93 Glu	19.52	0.065	1.97	300.08	280.64	0.61
94 Ala	23.58	0.079	2.55	299.58	281.41	4.10
95 His	18.60	0.062	1.31	300.69	273.12	0.28
97 Asp	23.31	0.078	2.52	299.02	280.85	0.95
98 Cys	22.59	0.076	2.59	297.67	280.51	0.26
99 Ile	22.05	0.074	2.58	298.23	281.41	0.11
101 Asp	22.34	0.074	2.42	300.23	282.09	1.08
104 Leu	29.54	0.098	3.12	300.93	282.34	0.78
105 Ser	27.75	0.092	3.15	300.05	282.72	1.18
107 Gly	19.28	0.065	2.44	296.58	281.00	3.58
108 Val	22.33	0.075	2.03	299.01	277.49	0.50
109 Met	26.64	0.089	3.25	299.41	283.26	0.63
110 Thr	33.20	0.111	3.79	300.20	282.97	0.23
112 Glu	28.52	0.094	2.88	301.86	282.43	0.41
113 Ile	17.44	0.059	2.11	297.57	281.29	0.12
115 Ala	22.32	0.075	4.01	297.04	286.00	2.48
116 Phe	23.03	0.077	3.09	299.15	284.44	0.62
117 Gly	27.62	0.093	3.48	298.20	282.56	0.98
119 Tyr	24.94	0.083	2.72	300.26	282.24	0.22
120 Val	15.70	0.053	1.68	297.29	278.94	0.33
127 Asn	23.00	0.077	2.46	296.97	278.61	5.81
128 Lys	15.54	0.052	1.35	300.36	277.87	0.66
129 Gln	14.27	0.048	1.80	296.82	281.19	0.20
130 Ile	27.87	0.093	3.19	300.92	283.73	0.02
131 Trp	21.74	0.073	2.54	298.55	281.71	0.04
132 Leu	26.43	0.088	3.33	300.68	285.03	0.02
133 Ala	29.63	0.099	3.27	300.17	282.36	0.19
134 Ser	20.31	0.068	2.07	299.70	280.45	0.13
136 Leu	13.22	0.044	1.27	299.25	278.86	0.25
140 Asn	25.41	0.085	2.80	299.43	281.59	0.17
142 Phe	20.94	0.070	3.06	299.27	285.74	0.03
143 Asp	21.86	0.073	2.50	300.24	283.04	0.13
146 Asn	23.64	0.080	3.61	295.03	282.08	2.00
147 Gly	25.22	0.085	2.37	297.81	276.98	4.80
148 Glu	21.59	0.072	2.69	298.80	282.98	1.40
150 Val	22.92	0.076	2.20	300.74	280.33	0.03
151 Ser	22.70	0.076	2.39	299.87	281.22	0.05
152 Leu	32.20	0.107	3.87	300.00	283.61	0.16
154 Asn	21.87	0.074	2.40	295.07	277.17	1.14
158 Leu	29.11	0.096	3.11	301.94	283.55	0.03
159 Thr	29.76	0.099	3.38	300.04	282.72	0.09
160 Asp	23.61	0.078	2.55	301.19	283.00	0.28
161 Ile	15.59	0.052	1.68	300.08	281.85	0.09
163 Thr	21.9	0.073	2.48	300.30	282.93	0.15
166 Val	27.28	0.091	2.81	300.84	281.79	0.06
168 Lys	17.28	0.058	1.93	299.93	282.33	0.16
170 Ile	22.63	0.075	2.11	301.25	280.25	0.31
172 Lys	28.06	0.095	3.84	294.06	279.64	1.5
174 Gln	20.58	0.069	2.20	297.53	279.16	
131 Trp e1	28.08	0.094	3.34	299.36	282.87	

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