Small-molecule inhibitors of the PDZ domain of Dishevelled proteins interrupt Wnt signalling

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Abstract

19 Dishevelled (Dvl) proteins are important regulators of the Wnt signalling pathway, interacting through 20 their PDZ domains with the Wnt receptor Frizzled. Blocking the Dvl PDZ/Frizzled interaction represents 21 a potential approach for cancer treatment, which stimulated the identification of small molecule 22 inhibitors, among them the anti-inflammatory drug Sulindac and Ky-02327. Aiming to develop tighter 23 binding compounds without side effects, we investigated structure-activity relationships of 24 sulfonamides. X-ray crystallography showed high complementarity of anthranilic acid derivatives in the 25 GLGF loop cavity and space for ligand growth towards the PDZ surface. Our best binding compound 26 inhibits Wnt signalling in a dose-dependent manner as demonstrated by TOP-GFP assays (IC₅₀ ~50 μM), 27 and Western blotting of β-catenin levels. Real-time PCR showed reduction in the expression of Wnt-28 specific genes. Our compound interacted with Dvl-1 PDZ (K_D=2.4 μM) stronger than Ky-02327 and

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may be developed into a lead compound interfering with the Wnt pathway.

INTRODUCTION

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35 Dishevelled (Dvl) proteins comprise 500 to 600 amino acids and contain three conserved domains: an 36 N-terminal DIX (Dishevelled/Axin) domain (Schwarz-Romond 2007, Madrzak 2015), a central PDZ 37 (PSD95/Dlg1/ZO-1) domain (Doyle 1996, Ponting 1997), and a C-terminal DEP (Dishevelled/Egl-38 10/Pleckstrin) domain (Wong 2000, Wallingford 2005). Dvl transduces Wnt signals from the membrane 39 receptor Frizzled to downstream components via the interaction between Dvl PDZ and Frizzled (Wong 40 2003), thus it has been proposed as drug target (Klaus 2008, Holland 2013, Polakis 2012). Several 41 studies identified internal peptides of the type (KTXXXW) as well as C-terminal peptides of the type 42 $(\Omega\Phi GWF)$ in which Ω is any aromatic amino acid (F, W or Y) as Dvl PDZ targets (Lee 2009, Zhang 43 2009). Three Dvl homologues, Dvl-1, Dvl-2 and Dvl-3, have been identified in humans. Sequence 44 identity is 88% between Dvl-3 PDZ and Dvl-1 PDZ and 96% between Dvl-3 PDZ and Dvl-2 PDZ 45 (Supporting Information Figure S1). Dvl proteins are found to be upregulated in breast, colon, prostate, 46 mesothelium, and lung cancers (Weeraratna 2002, Uematsu 2003, Bui 1997, Mizutani 2005). 47 PDZ domains appear in 440 copies spread over more than 260 proteins of the human proteome (Ponting 48 1997). They maintain relatively specific protein-protein interactions and are involved, for example, in 49 signalling pathways, membrane trafficking and in the formation of cell-cell junctions (Zhang 2003, 50 Fanning 1996, Kurakin 2007). Hence, they are potentially attractive drug targets (Rimbault 2019, 51 Christensen 2020). PDZ domains consist of about 90 amino acids which fold into two α-helices and six β-strands exposing a distinct peptide-binding groove (Doyle 1996), Lee 2017). The conserved 52 53 carboxylate-binding loop (GLGF loop, FLGI in Dvl-2 and -3, Figure 1) is essential for the formation of 54 a hydrogen bonding network between the PDZ domain and PDZ-binding, C-terminal peptide motifs, in 55 most cases coordinating the C-terminal carboxylate group of the interaction partner. In the respective 56 complexes, the C-terminal residue of the ligand is referred to as P₀; subsequent residues towards the N-57 terminus are termed P₋₁, P₋₂, and P₋₃ etc. Previous studies have revealed that P₀ and P₋₂ are most critical 58 for PDZ-ligand recognition (Songyang 1997, Schultz 1998). 59 PDZ domains are divided into at least three main classes on the basis of their amino acid preferences at 60 these two sites: class I PDZ domains recognize the motif S/T-X-Φ-COOH (Φ is a hydrophobic residue 61 and X any amino acid). Class II PDZ domains recognize the motif Φ-X-Φ-COOH, whereas class III

PDZ domains recognize the motif X-X-COOH. However, some PDZ domains do not fall into any of these specific classes (Pawson 2007, Sheng 2001, Zhang 2003). The Dvl PDZ domains, for example, recognize the internal sequence (KTXXXW) within the frizzled peptide 525(GKTLQSWRRFYH)536 $(K_D \sim 10 \mu M)$ (Wong 2003, Chandanamali 2009). Due to their occurrence in important proteins, PDZ domains received early attention as drug targets, nicely summarized in Christensen 2019. There are several examples of Dvl PDZ inhibitors of peptide or peptidomimetic nature (eg. Hammond 2006, Haugaard-Kedstrom 2021), including peptide conjugates (eg. Qin 2021, Hegedus 2021), and on an organic, small-molecule basis. The latter approach is considered most beneficial in long term medical treatments of conditions like cancer or neurological disorders. NSC668036 (Shan 2005, Wang 2015) is a peptide-mimic compound which interferes with Wnt signalling at the Dvl level. Based on a computational pharmacophore model of NCS668036, additional compounds were later reported (Shan 2012). Known as first non-peptide inhibitor, the 1Hindole-5-carboxylic acid derivative FJ9 (Fujii 2007) showed therapeutic potential. Further examples including Sulindac (Lee 2009), 2-((3-(2-Phenylacetyl)amino)benzoyl)amino)benzoic acid (3289-8625, also called CalBioChem(CBC)-322338) (Grandy 2009, Hori 2018), N-benzoyl-2-amino-benzoic acid analogs (Hori 2018), phenoxyacetic acid analogs (Choi 2016), and Ethyl 5-hydroxy-1-(2-oxo-2-((2-oxo-2-(1800)))). (piperidin-1-yl)ethyl)amino)ethyl)-1H-indole-2-carboxylate (KY-02327) (Kim 2016) have been reported, with the latter showing the highest in-vitro affinity (8.3 µM) of all. Despite the existence of the abovementioned inhibitors of Dvl PDZ, the development of tighter-binding, non-peptidic smallcompound modulators of the respective functions, binding with nanomolar affinity, is necessary and remains challenging. On this path, we explore optimal fits for the primary binding pocket by cycles of chemical synthesis and X-ray crystallography and further avenues for systematically growing ligands along the Dvl PDZ surface to provide SAR for the development of inhibitors in the low or medium nanomolar range. Nuclear magnetic resonance (NMR) spectroscopy was used to detect primary hits and for follow-up secondary screening. The ability of NMR to detect weak intermolecular interactions (µM < K_D< mM) make it an ideal screening tool for identifying and characterizing weakly binding fragments, to be optimized subsequently by chemical modification in order to improve binding (Zartler 2006, Shuker 1996, Zartler 2003). Besides NMR, the determination of X-ray crystal structures of selected

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complexes was fundamental for further design of new compound structures with improved binding. In the first round of screening, a library constructed after computational docking of candidates into the peptide binding site of the Dvl PDZ domains were investigated, followed by secondary screening utilizing a library of 120 compounds containing rhodanine or pyrrolidine-2,5-dione moieties.

RESULTS AND DISCUSSION

PDZ targeted library design

The PDZ targeted library was designed to cover all PDZ domains with available structure. For this, all X-ray and NMR derived PDZ structures were retrieved from the PDB, clustered, and 6 selected centroids were subjected to the virtual screening routine. The area considered is shown in Figure 1A, with the blue sphere indicating the geometrical centre. The clustering of the PDZ domains was performed according to the shapes of their binding sites, rather than backbone conformation. This approach accounts for the importance of surface complementarity of protein-small molecule interactions and the critical contribution of van der Waals interactions to the binding free energy. On another hand, PDZ domains have evolved to recognize a carboxyl group that is mostly derived from the C-terminus of natively binding proteins. Finally, the fact that PDZ can recognize internal motifs (Hillier 1999), including KTXXXW of Frizzled-7 recognised by Dvl PDZ (Wong 2003, Chandanamali 2009), raises the question of what are key binding contributions with PDZ domains: negative charge, hydrogen bonding or shape complementarity (Harris 2003). For this reason, tangible compounds were preselected to have extensive hydrophobic contacts as well as chemical groups that mimic the carboxylic group.

Virtual screening was performed with QXP, and the generated complexes were sequentially filtered with

NMR Screening and development of compounds

250 compounds was subjected to NMR validation.

The results of virtual screening were checked experimentally by comparing 2D ¹H-¹⁵N HSQC (Heteronuclear Single Quantum Correlation) spectra of Dvl-3 PDZ in the absence and presence of the compound to elucidate ligand-induced changes of chemical shifts. Chemical shift perturbation

a self-designed MultiFilter algorithm. From the resulting 1119 compounds a randomly selected set of

differences (Δ CSP, representing the average of the three strongest shifting cross peaks according to equation 1) were evaluated in cases where the residues responding strongest are inside the area defined by Figure 1A. The responses were classified into: (i) inactive compounds (Δ CSP < 0.02); (ii) very weak interactions ($0.02 \le \Delta$ CSP ≤ 0.05); (iii) weak interactions ($0.05 < \Delta$ CSP ≤ 0.1); (iv) intermediate interactions ($0.1 < \Delta$ CSP ≤ 0.2); (v): strong interactions ($0.2 < \Delta$ CSP ≤ 0.5) and (vi) very strong interactions (Δ CSP > 0.5). In most cases, the signals of residues S263, V287 and R320 (Figure 1A) within the conserved binding site were most strongly perturbed (Supporting Information Figure S2). With the Δ CSP of 0.12 ppm, the isoleucine-derived compound 1 ((2,3-dihydrobenzo[b][1,4]dioxin-6-yl)sulfonyl)-L-isoleucine containing a sulfonamide moiety was detected initially as one of the best "hits" according to chemical shift changes. The sulfonamide is a well-known moiety in drug discovery (Mathvink 1999, Wu 1999, Sleight 1998 O'Brien 2000, Tellew 2003).

Upon NMR titration experiments for compound 1 (Supporting Information Figure S2) with Dvl-3 PDZ, the largest chemical shift perturbations were observed for S263 in strand β B and R320 in helix α B of Dvl-3 PDZ, confirming the conserved binding site.

136 Scheme 1: Compounds 2, 3, 4, 5

By comparing the binding of several sulfonamide compounds in a secondary screening event and making use of our in-house library, four new compounds (2, 3, 4, 5) that induced chemical shift perturbations larger than 0.2 ppm were found (for binding constants see Table 1) and considered further as reasonably

strong binders. The similarity of the structures led us to define Scheme 2 as a scaffold for further refinements. Sulfonamides were considered more drug-like, and hence followed up at higher priority than other hits. We realised that our four new compounds had different moieties at R_2 in combination with a small R_1 (fluorine). A decrease of binding was observed with decreasing size of R_2 .

Scheme 2: Basic fragment for further synthesis

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In order to assess the importance of the aryl group at R₂ for complex formation, it was replaced by a methyl group as substituent to yield compound 6, which showed a drastic decrease of binding (Table 1). Compounds 3, 4, and 5 did not distinguish between the Dvl-3 PDZ and Dvl-1 PDZ. In order to obtain detailed insight into the binding mode of these compounds, crystal structures of Dvl-3 PDZ in complex with compounds 3, 5 and 6 were determined (Figure 1). For compound 3 the crystal structure revealed two complexes within the crystallographic asymmetric unit (AU) at 1.43 Å resolution. Both show the anthranilic acid with the attached fluorine pointing into the hydrophobic binding pocket (Figure 1B and Supporting Information Figure S3A), while the carboxyl group forms a hydrogen-bond network with amide residues of the carboxylate binding loop, in particular strand βB (Figure 1B) and specifically with residues I262, G261 and L260. The two sulfonamide oxygen atoms form hydrogen bonds with R320 and H324 (weak) of helix αB for only one complex in the AU. The aromatic aryl group (tetrahydronaphtalene) attached to the sulfonamide is involved in hydrophobic interactions with F259 (Supporting Information Figure S3B). The 1.6-Å complex structure with compound 5 (4 molecules per AU) exhibits a comparable binding mode as found for compound 3 with a hydrogen-bond network involving the carboxyl group and the amides of I262, G261, L260, and of the sulfonamide to H324 (Figure 1C). No hydrogen bond was observed to R320 in all four molecules of the AU, but small variations of the aryl moiety relative to F259 (Supporting Information Figure S3C). The crystals of the complex with 6 show two molecules in the AU (Figure 1D). The sulfonamide is bound by H324 in both

	ID	\mathbf{R}_1	R_2	(K _D , μM) Dvl-3PDZ	(K _D , μM) Dvl-1 PDZ
	2	F		nd	237.6 ± 38.5^{NMR}
	3 4	F F		$80.6 \pm 6.1^{NMR} \\ 83.9 \pm 7.8^{NMR}$	112.7 ± 25.9^{NMR} 114.4 ± 9.8^{NMR}
	5	F		140.6 ± 14.1^{NMR}	160.1 ± 14.6^{NMR}
	6	F	CH ₃	> 1000 ^{ITC}	-
	7	Br	△	20.6 ± 2.4^{NMR}	$18.2 \pm 2.4^{\text{NMR}}$
O O O NH O	8	CF ₃	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	$17.4 \pm 0.5^{\rm ITC}$	$24.5\pm1.5^{\mathrm{ITC}}$
ОН	9	Cl		41.1 ± 3.1^{NMR}	45.6 ± 4.5^{NMR}
R ₁	10	CH ₃		62.5 ± 4.7^{NMR}	60.5 ± 5.3^{NMR}
-4	11	Br		13.8 ITC	119.9 ^{ITC}
	12	Br		58.5 ITC	nd
	13	Br		7.2 ITC	213.2 ^{ITC}
	14	Br		$58.1 \pm 2.1^{\rm ITC}$	nd
	15	CF ₃		$52.9 \pm 1.7^{~\rm ITC}$	nd
	16	CF ₃		59.1 ± 1.5^{ITC}	nd
	17	CF ₃	Ů,	49.5 ^{ITC}	nd
	18	CH ₃	N H	$9.4 \pm 0.6 \ ^{\mathrm{ITC}}$	$2.4 \pm 0.2^{\rm ITC}$
	19	CH ₃	HN-IN CI	$21.8 \pm 1.7^{\rm ITC}$	$8.0 \pm 0.5^{\rm ITC}$
	20	CH ₃	Br NH H	$9.8 \pm 0.3^{\rm ITC}$	$4.7 \pm 0.3^{\rm ITC}$
	21	CH ₃	CI NH H	$12.5 \pm 0.5^{\text{ITC}}$	$4.7 \pm 0.2^{\rm ITC}$

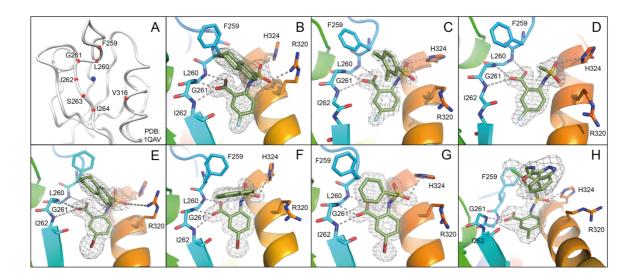


Figure 1: A Definition of PDZ binding site. The center of the binding site (blue sphere) is defined as the geometric center of Cα atoms (red spheres) of 7 residues (typed in red) defined by multiple sequence alignment. B-H Magnified views into crystal structures of various compounds bound to the Dvl-3 PDZ domain. The 2Fo-Fc electron density around the compounds is shown at 1σ contour level, and the dotted lines indicate formed hydrogen bonds. In the bound compounds covalent bonds to carbon atoms are shown as green sticks. Important residues involved in compound binding are labelled and displayed in atom colours (carbons blue or dark yellow). B-D show compound 3, 5 and 6 respectively. All compounds in B-D contain fluorine (light blue) in para position to the amine. E-G represents the bound compounds 7, 11 and 12, respectively. All have bromine (dark red) in para position to the amine. H shows compound 18 within the binding site. The accession codes of the structures B-H are 6ZBQ, 6ZBZ, 6ZC3, 6ZC4, 6ZC6, 6ZC7 and 6ZC8, respectively.

To further explore the importance of the fluorine site inside the hydrophobic pocket, substitutions by bromine, chlorine, methyl and trifluoromethyl were chosen. In fact, the methyl group has a similar vdW radius as the CF₃ group. Iodine was not considered a good candidate since it increases molecular weight substantially and the compounds may be chemically less stable, in particular in biological assays. Taking into account that compound 6 did not bind because of the missing aromatic ring at the R₂ position, our overall strategy was to increase the aromatic ring at R₂ while finding a good fit for R₁, keeping an eye on the molecular weight to enable further compound modifications that fulfil key properties as defined by Lipinski (Lipinski 2000, Lipinski 1997). Our preference to continue exploration at the R₁ position of the aromatic ring in Scheme 1 was inspired by the absence of hits with other substitutions in the secondary screening event and the initial X-ray structures that showed a hydrophobic pocket available for substituents in this position while other sites at the aromatic ring would include steric hindrance. Therefore, compounds 7-17 were obtained and were classified in three different groups to derive structure activity relationships (SAR). The compounds 7-10 in group 1 contain different R₁ (Br, CF₃, Cl, CH₃) but the same moiety (tetrahydronaphthalene) at R₂. As expected, binding could be further improved by displacement of the fluorine with elements exhibiting larger van der Waals (vdW) radii.

199 Indeed, the K_D decreased stepwise and the best fit was observed for compound 8 containing a 200 trifluoromethyl group ($K_D = 17.4 \mu M$ for Dvl-3 PDZ and 24.5 μM for Dvl-1 PDZ). The different 201 substituents at the R₁ position contribute to an increased binding affinity in the following order: F < Cl 202 < Br < CF₃ (compound 3 < 9 < 7 < 8, respectively). Compound 10 with a methyl group at the R₁ position 203 showed only marginally improved binding, although the methyl group has a similar vdW radius as the 204 CF₃ group of compound 8. The difference in binding results most likely from their different 205 hydrophobicity. The 1.85-Å crystal structure of the Dvl-3 PDZ domain with compound 7 (K_D = 20.6 μM for Dvl-3 PDZ 206 207 and 18.2 µM for Dvl-1 PDZ) showed an identical hydrogen-bond network involving the amide groups 208 of residues I262, G261 and L260 of the carboxyl binding loop as seen for all other complex structures 209 reported here (Figure 1E). Only one hydrogen bond between the sulfonamide and R320 was found in 210 addition for one of the two Dvl-3 PDZ molecules per AU. H324 of Dvl-3 PDZ was not addressed by the 211 sulfonamide as seen previously. The bromine at position R₁ points into the hydrophobic pocket, similar 212 as the fluorine in the complex structure with compound 3. The two complexes in the AU show significant 213 variations in the positions of the tetrahydronaphtalene rings as well as for the side chain of F259 and 214 R320 (Supporting Information Figure S3E). 215 Following the analysis of the complex involving compound 7, the binding characteristics of the group-216 2 compounds (11-14) were investigated. They contain bromine as R₁ and different substituents at the R₂ 217 position to assess the importance of π - π stacking interactions involving F259. K_D values of 7.2 μ M for 218 compound 13 and 13.8 µM for compound 11 were found with respect to the interaction with Dvl-3 PDZ. 219 Crystal structures of Dvl-3 PDZ in complex with compound 11 (1.58 Å resolution, 1 molecule per AU) 220 and 12 (1.48 Å, 2 molecules per AU) revealed very similar binding as observed in the crystal structures 221 with compounds 3 and 7. The aromatic rings at R_2 show hydrophobic interactions to F259, but not a 222 classical π - π stacking as expected. Nevertheless, the tighter binding of compound 11 could be explained 223 by the larger aromatic substituent at the R_2 position compared to compound 12. Both complex structures 224 show also non-specifically bound ligands in crystal contacts (Supporting Information Figure S3H, 225 Supporting Information Tables S2 and S3). The additional ligand molecules in both complex structures 226 can be explained as a crystallographic artefact, which is verified with ITC experiments that indicate 1:1 stoichiometries in both cases (Figure S5). With respect to the selectivity of the tested compounds we observed a 6 to 30-fold stronger binding of compounds 7, 9, 11 and 13 to Dvl-3 PDZ as compared to Dvl-1 PDZ. These differences are related to the different sequences at the end of αB. Most importantly, H324 is replaced by a serine residue in the Dvl-1 PDZ domain.

The group-3 compounds (15-17) contain a trifluoromethyl at position R₁ and were tested to investigate

The group-3 compounds (15-17) contain a trifluoromethyl at position R_1 and were tested to investigate a cooperative role of this moiety with various substituents at position R_2 . All compounds bind weaker to Dvl-1 and Dvl-3 than compound 8 which contains tetrahydronaphthalene at the R_1 position, revealing its important role in the interaction.

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Further modifications towards higher affinity and reduced toxicity

Possible cytotoxic effects of compounds 3, 7, 8, 9, and 10 were evaluated in cell viability assays using HEK293 cells (Supporting Information Figure S4). These compounds were selected due to different substituents at the R₁, including halogens. Cell viability was measured 24h after treatment with the individual compounds, and half maximal inhibitory concentrations (EC₅₀) were calculated for each compound. The compounds exhibited EC₅₀ values in the range of 61-131 µM (Supporting Information Figure S4A). Compounds 3 and 10 that contained fluorine or methyl group substituents at R₂, respectively, were the least toxic, while compound 7, containing bromine, was the most toxic. The results from crystallography, modelling studies and of the cell proliferation assays led us to further investigate compounds 18-21 that contain a methyl group at the R₁ position and different substituents as R₂. In this way, we aimed to develop both potent and less toxic, cell permeable inhibitors. All compounds showed strong interactions as indicated by chemical shift perturbation values between 0.30 to 0.34 ppm (Supporting Information Table S1). The binding constants were evaluated by ITC whereby compound 18 ($K_D = 9.4 \mu M$ for Dvl-3 PDZ and 2.4 μM for Dvl-1 PDZ) appeared to be most potent. Compound 18 contains a pyrazole ring which is considered as an important biologically active heterocyclic moiety (Lv 2010). Compounds 20 ($K_D = 9.8 \mu M$ for Dvl-3 PDZ and 4.7 μM for Dvl-1 PDZ) and 21 ($K_D = 12.5 \mu M$ for Dvl-3 PDZ and 4.7 μM for Dvl-1 PDZ) contain pyrrole rings. Their binding constants almost have the same value despite the different substituents (bromine or chlorine) at the pyrrole rings. The binding of compounds 18-21 to both Dvl PDZ domains is mainly enthalpy-driven as indicated in Table 2, with a slightly stronger effect for Dvl-1 PDZ than for Dvl-3 PDZ. To our surprise, the crystal structure of Dvl-3 PDZ in complex with compound 18 shows the pyrazole substituent in the R_2 position orientated away from the binding pocket. Instead, a π - π stacking interaction with F259 was observed (Supporting Information Figure S3I). Cytotoxicity of 18-21 was determined *via* MTT assays (Mosmann 1983) that displayed viability up to concentrations above 150 μ M (Supporting Information Figure S4B).

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Compound	Dvl-3 PDZ		Dvl-1 PDZ						
	$K_{D}\left(\mu M\right)$	ΔH	$T\Delta S$	ΔG	$K_D(\mu M)$	ΔH	$T\Delta S$	ΔG	
		(kcal/	(kcal/	(kcal/		(kcal/	(kcal/	(kcal/	
		mol)	mol)	mol)		mol)	mol)	mol)	
18	9.4 ± 0.6	-8.0	-1.2	-6.8	2.4 ± 0.2	-12.2	-4.7	-7.5	
19	21.0 ± 1.7	-5.9	0.4	-5.5	8.0 ± 0.5	-7.3	-0.3	-7.0	
20	9.8 ± 0.3	-10.4	-3.6	-6.8	4.7 ± 0.3	-9.4	-2.2	-7.2	
21	12.5 ± 0.5	-5.9	0.7	-6.8	4.7 ± 0.2	-8.5	-1.5	-7.0	
NPL-1011	79.7 ± 53.3								
Sulindac	8.3 ± 2.5								
CBC-322338/									
3289-8625	$>$ 400 μM								
NSC668036	$> 400 \mu M$								
Ky-02327	$8.3\pm0.8^{16\mathrm{g}}$								

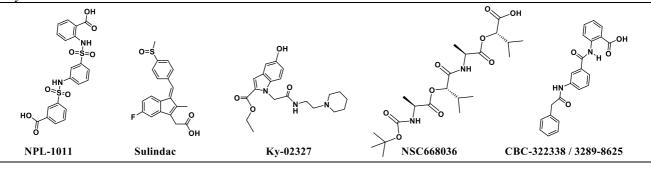


Table 2: Isothermal titration calorimetric data for the reaction between Dvl-3 PDZ, Dvl-1 PDZ and our compounds **18, 19, 20** and **21** respectively. Compounds NPL-1011 (Hori 2018), and Sulindac (Lee 2009), CBC-322338/3289-8625 (Grandy 2009, Hori 2018), and NSC668036 (Shan 2005), for more thermodynamic parameters see Supporting Information Figure S7. For Ky-02327 the value from literature is included.

Comparison to reported Dvl PDZ-binding molecules

Our compounds bind to Dvl-3 with a K_D better than 10 μ M, and slightly tighter to Dvl-1, see Table 2, with 18 showing a K_D of 2.4 μ M and chemical shift changes indicating binding to the canonical binding site (Figure 1A). For comparison, four compounds of those shown in Supporting Information Figure S6 were assayed by ITC (Supporting Information Figure S7) regarding their affinity to Dvl-3 PDZ. Ky-02327 was already determined to bind with a K_D of 8.3 ± 0.8 μ M (Kim 2016) to Dvl-1 PDZ. Our first interest was oriented towards sulfonamides. Hori et al (Hori 2018) have recently reported 3-({3-[(2-carboxyphenyl)sulfamoyl]phenyl}sulfamoyl)benzoic acid (NPL-1011) binding to Dvl-1 PDZ via the

detection of chemical shift changes, and further sulfonamide compounds that showed smaller effects, indicating weaker binding. We examined the binding constant of NPL-1011 which possesses two sulfonamide moieties by ITC and found a value of $79.7 \pm 53.3 \,\mu\text{M}$, see Table 2. For further comparisons, we assayed also CBC-322338/3289-8625, Sulindac and NSC668036 by ITC. Surprisingly, CBC-322338/3289-8625 showed very low affinity, with a K_D above 400 μ M in our ITC assay, in line with the value found by Hori et al (Hori 2018) (954 +/- 403 μ M). We also applied an NMR shift assay (Figure S8), yielding a Δ CSP around 0.1. Based on NMR and ITC studies, the binding affinity of CBC-322338/3289-8625 to DVL-3 seems to be less than 50 micromolar (comparing the CSPs from the NMR assay with those of our other compounds listed in Table S1 and the respective binding constants in Table 1, and considering also the weak heat development in our ITC assay) which was larger than the originally reported value (10.6 +/- 1.7) (Grandy 2009) that was obtained with a different method. Concerning non-sulfonamide compounds, a K_D of 8.3 \pm 2.5 μ M was detected for Sulindac, while NSC668036 (Shan 2005) did not show high-affinity binding. These results are largely in agreement with literature. In all cases, compounds were tested for purity after K_D measurements (see Supporting Information Figures S9A-D).

Selectivity testing using a set of selected PDZ domains

Compounds 18, 20 and 21 were tested towards other PDZ domains for selectivity. The set included PSD95-PDZ 2 and 3, Shank-3, α -syntrophin, and AF-6 PDZ. According to the determined chemical shift perturbations (Supporting Information Table S4), our compounds show no or very weak interactions with the selected PDZ domains (0.05 < Δ CSP \leq 0.1) ppm. These findings led to the conclusion that our compounds show considerable selectivity towards Dvl PDZ domains. This selectivity might be due to a unique feature of Dvl PDZ where R320 (Dvl-3 PDZ) or R322 (Dvl-1 PDZ) are crucial for interactions, explaining selectivity with respect to other PDZ domains. In addition, the large hydrophobic cavity for the side chain of the C-terminal residue of the interacting peptide is occupied by a large moiety in case of compounds 18, 20 and 21 which might not be accommodated in most other PDZ domains.

Taking advantage of a lentivirus that encodes GFP in a β-catenin/TCF-dependent fashion (TOP-GFP, SABiosciences), a stable HEK293 reporter cell line was established to evaluate the inhibitory effect of compounds 18, 20 and 21 on canonical Wnt signalling activity. TOP-GFP expression in this cell line was induced by the ligand Wnt3a, which directly activates the Frizzled-Dishevelled complex and protects β-catenin from degradation by the destruction complex (Figure 2A). Remarkably, all three compounds inhibited Wnt signalling induced by Wnt3a in a dose-dependent manner (Figure 2B), yielding IC₅₀ values between 50-80 μM.

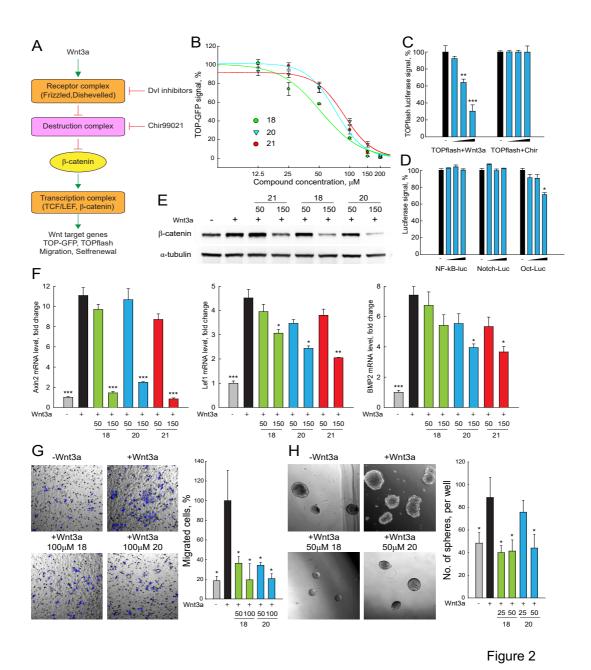


Figure 2. DVL inhibitors antagonize Wnt signalling and Wnt related properties of cancer cells induced by Wnt3a. A. Scheme of Wnt signalling pathway. Important components of the Wnt signalling pathway are schematically presented. Wnt3a treatment

increases the transcription of Wnt targets, enhances signals of TOP-GFP and TOPflash assays, and promotes Wnt related biological properties of cancer cells. **B.** TOP-GFP reporter assays were performed with HEK293 reporter cell line. Compound **18**, **20** and **21** inhibited Wnt3a induced Wnt activation in dose-dependent manner with IC50 of $50\sim75~\mu M$. **C&D**. TOPflash assays stimulated with Chir99021 and reporter assays of other pathway were used to evaluate the specificity of compound **20**. Compound **20** specifically inhibited Wnt3a induced Wnt activation, and had no or mild effect on Chir99021 induced Wnt activation and other signalling pathways including NF- κ B, Notch and Oct4. **E.** β -Catenin protein levels were detected with Western blotting in Hela cells. Compound **18**, **20** and **21** (150 μ M) inhibited accumulation of β -catenin in Hela cells treated with Wnt3a. **F.** The mRNA levels of Wnt target genes (Axin2, Lef1 and Bmp2) in Hela cells were measured with quantitative real-time PCR. Compounds **18**, **20** and **21** (150 μ M) reduced the transcription of Wnt target genes that are enhanced by Wnt3a treatment in Hela cells. **G.** Cell migration of SW480 cells after Wnt3a treatment was assessed by transwell assays. Compounds **18** and **20** (50~100 μ M) reduced the migration of SW480 cells enhanced by Wnt3a. **H.** SW480 cells were cultured in serum-free non-adherent condition to evaluate the self-renewal property enhanced by Wnt3a treatment. Compound **18** and **20** (25~50 μ M) reduced sphere formation of SW480 cells that was enhanced by Wnt3a treatment. For all tests, three independent biological replicas were performed and error bars represent standard deviations. P-values were calculated from T-test. *: P < 0.05; **: P < 0.01; ***: P < 0.001.

To further evaluate the specificity of our Dvl inhibitors, the conventional TOPflash (Molenaar 1996)

> and other luciferase reporter assays were performed. In Hela cells, 20 inhibited TOP-luciferase signals stimulated by Wnt3a but not by CHIR99021(Sineva 2010), a compound that activates Wnt signalling downstream of Dvl (Figure 2A, C). Compound 20 had no significant inhibitory effects in reporter assays that measure the activity of other signalling systems, e.g., NF-κB-luciferase stimulated by recombinant TNFα, Notch-luciferase stimulated by the overexpression of the Notch intracellular domain, or the Octluciferase assay that is stimulated by overexpression of Oct4 (SABiosciences, Figure 2D). These results strongly indicate that 20 is specific for canonical Wnt signalling at the upstream level. Increased β-catenin protein level is a hallmark of active Wnt signalling (Kishida 1999). Once β-catenin is accumulated in the cytoplasm, it can translocate into the nucleus and activate the transcription of Wnt target genes by interacting with transcription factors of the TCF/LEF family (Figure 2A) (Behrens 1996). In Hela cells, all three Dvl inhibitors blocked the increase of production of β-catenin by Wnt3a in dose-dependent manners, as seen by Western blotting (Figure 2E). Increased mRNA levels of the Wnt target genes Axin2, Lef1 and Bmp2 (Riese 1997, Jho 2002, Lewis 2010) were induced by Wnt3a treatment, as measured by qRT-PCR, and these increases were reduced by compounds 18, 20 and 21 (Figure 2F). These results demonstrate that compounds 18, 20 and 21 inhibit Wnt signalling as indicated by reduced accumulation of β -catenin and low expression of typical Wnt target genes. Canonical Wnt signalling contributes to cancer progression by inducing high motility and invasion of cancer cells while retaining the self-renewal property of cancer initiating cells (Fritzmann 2009, Sack 2011, Vermeulen 2010, Malanchi 2008). In particular, cancer initiating cells are propagated and enriched in non-adherent sphere culture, demonstrating the self-renewal capacity of the stem cells

(Kanwar 2010, Fan 2011). To investigate the potential value of the Dvl inhibitors for interfering with these Wnt-related properties of cancer cells, the subline SW480WL was derived from the SW480 colon cancer cell line, which exhibits a low level of endogenous Wnt activity (Fang 2012). The cell migration and self-renewal properties of SW480WL cells were enhanced by Wnt3a treatment, as revealed by transwell and sphere formation assays (Figure 2G, H). Compounds 18 and 20 prevented increased cell migration and sphere formation. These results indicate that our Dvl inhibitors may be developed into lead compounds that interfere with Wnt signalling.

CONCLUSIONS

In the present work, small molecules that bind to Dvl PDZ in the one-digit micromolar range with considerable selectivity have been developed by an extensive structure-based design approach. With regards to the affinity determined by ITC, compound 18 binds to Dvl-1 and Dvl-3 in vitro with K_D values of 2.4 and 9.4, respectively, comparing very well with known ligands. X-ray structures of Dvl-3 PDZ complexes with selected compounds provided insight into crucial interactions and served as the basis for the design of tight binding compounds with reduced toxicity. The structural investigations revealed that these compounds form hydrogen bonds with the amide groups of residues L260, G261 and I262 in the PDZ-domain loop and the side chains of residues H324 and R320. Finally, the chosen methodology, virtual screening followed by a two-stage NMR based screening, X-ray crystallography, and chemical synthesis is an excellent path towards bioactive interaction partners. Our best compounds effectively inhibited the canonical Wnt signalling pathway in a selective manner and could be developed for further studies.

Experimental Section

Clustering binding sites and selection of representative PDZ domains

Three-dimensional structures of PDZ domains were retrieved from the PDB (Berman 2000). At the time of the study from a total of 266 PDB files, 126 were NMR solution structures and 140 derived from X-ray diffraction studies. The structures belong to 163 PDZ domains of 117 different proteins from 11 organisms. Files which contain more than one 3D conformation for a domain (up to 50 for NMR-derived

data) were split into separate structures and considered independently. The total number of unique 3D structures was 2,708.

Amino acid sequences of PDZ domains were aligned using Clustal Omega software (Sievers 2011).

Based on the alignment, for each structure, residues which form the binding site (strand βB and helix αB) were determined (Supporting Information Figure S8). The centre of the binding site was defined as a geometric centre of $C\alpha$ atoms of 7 residues (6 residues from the βB strand and the second residue from the αB helix). Such bias toward the βB strand was made to cover sites occupied by residues in -1 and -

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392 The triangulated solvent accessible surface for each PDZ structure was built using MSMS software 393 (Sanner 1996) with a spherical probe radius of 1.4 Å and vertex density 10 Å⁻¹. The largest connected 394 set of surface vertices within 9 Å from the centre of the binding site was used to construct shape-based 395 numerical descriptors. The descriptors are 508-dimensional vectors of non-negative integer numbers 396 and were built using a shape distributions approach (Osada 2002). In total 10 (Pawson 2007) vertex 397 triplets were selected randomly, each forming a triangle. Triangles which had a side longer than 16 Å 398 were discarded. Triangle sides were distributed into 16 length bins, each 1 Å wide, covering lengths 399 from 0 to 16 Å. A combination of three sorted side lengths, each belonging to one of 16 distance bins, 400 defines one of 508 categories of the triangles. The number of triangles of each category was calculated, 401 resulting in a 508-dimensional vector which is used as a numerical descriptor of the binding pocket 402 shape. For further operations with descriptors, Euclidian metric was introduced. Shape descriptors were 403 distributed into 6 clusters using k-means algorithm (Jain 1988). For each cluster, a centroid structure 404 was defined as the one, whose descriptor is the closest to mean descriptor for the cluster. The centroid 405 structures (2O2T#B.pdb, 1VA8#3.pdb, 2DLU#01.pdb, 1UHP#8.pdb, 2OS6#8.pdb, 3LNX#A.pdb) were 406 used for docking.

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PDZ targeted library design

Screening collection by Enamine Ltd. (Chuprina 2010) containing a total of 1,195,395 drug-like compounds was used as the primary source of small molecules. Natural ligand of PDZ is the C-terminus of a peptide with carboxylic group making extensive hydrogen bond network with the " $\Phi G\Phi$ " motif.

Since the carboxyl provides either of negative charge and hydrogen bond acceptor, we want our ligands to retain at least one of these features. Therefore, we pre-filtered the stock library to bear chemical groups which have negatively charged and/or hydrogen bond acceptor functionality. In total 65,288 compounds were selected for the virtual screening study. The selected 6 centroids of PDZ domains and the prepared compound set were subjected to high-throughput docking using the QXP/Flo software (McMartin 1997). Complexes were generated with 100 steps of sdock + routine, and 10 conformations per complex were saved. Processing of docking poses started with filtering by contact term *Cntc* from the QXP/Flo scoring function. Entries with Cntc < 45 were discarded, which removed complexes with weak geometries of bound ligands. The remained filtering was performed with the in-house MultiFilter program that allows flexible geometry-based filtering. We applied two algorithms, nearest-atom filter and hydrogen-bond filter. The former filters complexes by distance from a given protein atom to the nearest heavy ligand atom, while in the latter, filtering is based upon the number of hydrogen bonds calculated for a given complex geometry. With the nearest-atom routine we selected compounds that filled the P₀ pocket and sterically mimicked binding of a peptide carboxylic group. Peptide group hydrogens of the "ΦGΦ" motif and atoms forming the hydrophobic pocket were used for that. With the hydrogen-bond filter we selected compounds that formed extensive hydrogen bonding with the PDZ domain. Both these properties might have larger impact on binding rather than negative charge (Harris 2003). Details on atoms used for filtering and thresholds for hydrogen-bond filtering, as well as the resulting number of compounds, are provided in Supporting Information Table S5. Compounds from complexes which passed through these filters were incorporated into a targeted library for the PDZ-domain family. The final library contained

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Screening of compounds

1119 compounds in total.

Two-dimensional $^{1}\text{H-}^{15}\text{N}$ HSQC spectra were used to screen a library of 212 compounds designed by the company Enamine for PDZ domains. 50 μM of $^{15}\text{N-labeled}$ protein samples were prepared in a 20 mM sodium phosphate buffer, containing 50 mM sodium chloride, 0.02% (w/v) NaN₃, at pH 7.4. Stock solutions of small molecules were prepared in DMSO-*d6* at a concentration of 160 mM. A $^{1}\text{H-}^{15}\text{N}$ HSQC

spectrum of Dvl PDZ was acquired at 300 K with 5% DMSO-d6 in the absence of ligand as reference spectrum. Mixtures of 16 compounds were added to ¹⁵N-labeled Dvl PDZ at 8-fold molar excess each. The final concentration of DMSO-d6 in the protein-ligand solutions was 5%. Spectra were acquired with 8 scans and 256 points in the indirect dimension. Compound binding was deduced if the resonance position of a cross-peak was significantly shifted compared to the reference spectrum. The active compound was obtained through successive deconvolution. Experiments were recorded on a Bruker DRX600 spectrometer equipped with a triple-resonance cryoprobe. The preparation of samples was done automatically by a Tecan Genesis RSP 150 pipetting robot. Spectra were analysed using the programs TOPSPIN and SPARKY.⁴⁷

Synthesis of compounds

All reagents and starting materials were purchased from Sigma-Aldrich Chemie GmbH, ABCR GmbH & Co.KG, alfa Aesar GmbH & Co.KG or Acros Organics and used without further purification. All air or moisture-sensitive reactions were carried out under dry nitrogen using standard Schlenk techniques. Solvents were removed by evaporation on a Heidolph Laborota 4000 with vacuum provided by a PC 3001 Vaccubrand pump. Thin-layer chromatography (TLC) was performed on plastic-backed plates precoated with silica gel 60 F₂₅₄ (0.2 mm). Visualization was achieved under an ultraviolet (UV) lamp (254 and 366 nm). Flash chromatography was performed using J.T Baker silica gel 60 (30-63 µm). Analytical high-performance liquid chromatography (HPLC) was performed on a Shimadzu LC-20 (degasser DGU-20A3, controller CBM-20A, autosampler SIL-20A) with a DAD-UV detector (SPD-M20A), using a reverse-phase C18 column (Nucleodur 100-5, 5 µM, 250 mm x 4 mm, Macherey-Nagel, Düren, Germany). Separation of compounds by preparative HPLC was performed on a Shimadzu LC-8A system equipped with a UV detector (SPD-M20A), using a semi-preparative C18 column (Nucleodur 100-5, 5 μM, 250 mm x 10 mm, Macherey-Nagel) or preparative C18 column (Nucleodur 100-5, 5 μM, 250 mm x 21 mm, Macherey-Nagel). The detection wavelength was 254 nm. Gradients of acetonitrilewater with 0.1% TFA were used for elution at flow rates of 1 mL/min, 8 mL/min, and 14 mL/min on the analytical, semi-preparative and preparative columns respectively. Melting points (mp) were determined with Stuart Melting Point Apparatus SMP3 and are not corrected. Mass spectra were recorded on a 4000Q TRAP LC/MS/MS/ System for AB Applied Biosystems MDS SCIEX. NMR spectra were recorded on a Bruker AV300 spectrometer instrument operating at 300 MHz for proton frequency using DMSO-d6 solutions. Chemical shifts were quoted relative to the residual DMSO peak (1 H: $\delta = 2.50$ ppm, 13 C: $\delta = 39.52$ ppm). Coupling constants (J) are given in Hertz (Hz). Splitting patterns are indicated as follows: singlet (s), doublet (d), triplet (t), quartet (q), multiple (m), broad (b). Purity of each compound used for biological testing was $\geq 95\%$ unless otherwise noted. The purity check of known inhibitors purchased for comparison with our compounds are found in Supporting Information Figure S9.

Synthesis of compounds 8, 11 – 17

$$R_1 = CF_3$$
, Br, Cl

$$R_2 = \bigcap_{i \in \mathcal{A}_i} \bigcap_{j \in \mathcal{A}_i$$

Scheme 3: Synthesis of compounds 8, 11 - 17

To a solution of anthranilic acid substituted with the appropriate R_1 (1.32 mmol) and sodium carbonate (3.17 mmol) in water (2 mL) at 80 °C, the sulfonyl chloride (1.58 mmol) substituted with the appropriate R_2 was added over a period of 5 minutes. The stirring continued for 18 h at 80 °C. The reaction mixture was cooled to room temperature and acidified with 6 N HCl, and the resulting solid precipitate was filtered, washed with water and dried to give the crude product. The final product was obtained by preparative HPLC (Puranik 2008).

2-(5,6,7,8-tetrahydronaphthalene-2-sulfonamido)- 5- (trifluoromethyl) benzoic acid (8)

- 489 (0.52 g, 74% yield) ¹**H-NMR** (300 MHz, DMSO-d6): $\delta = 11.77$ [s, 1H, COOH], 8.13 [s, 1H, NH],
- 490 7.85 [d, ${}^{3}J_{6,4}$ = 2.1 Hz, 1H, 6-H_{Ar}] 7.62 [d, ${}^{4}J_{1',3'}$ = 2.1Hz, 1H, 1'-H_{Ar}] 7.53 [dd, ${}^{3}J_{4,3}$ = 7.1 Hz, ${}^{4}J_{4,6}$ =
- 491 2.1 Hz, 4-H_{Ar}] 7.36 [dd, ${}^{3}J_{3'4'} = 7.5$ Hz, ${}^{4}J_{3',1'} = 2.4$ Hz, 1H, 3'-H_{Ar}] 7.15 [d, ${}^{3}J_{4',3'} = 7.5$ Hz, 1H,4'-H_{Ar}]
- 492 , 6.90 [d, ${}^{3}J_{3,4}$ = 7.1Hz, 1H, 3-H_{Ar}] 2.73 (m, 4H, CH₂); 1.6 (m, 4H, CH₂); -13C-NMR (75 MHz, DMSO-
- 493 d6): $\delta = 169.1(C, C_{Ar}-8], 152.7(C, C_{Ar}-2), 143.8(C, C_{Ar}-4a'), 138.7(C, C_{Ar}-2'), 135.9(C, C_{Ar}-8a'),$
- 494 130.4(CH, C_{Ar}-4), 128.7 (CH, C_{Ar}-6), 127.5 (CH, C_{Ar}-1'), 124.0 (CH, C_{Ar}-4'), 121.6 (C, C-6), 118.2 (C,
- 495 C_{Ar}-5), 116.9 (C, C_{Ar}-3), 29.0 (CH₂, C-8'), 28.8 (CH₂, C-5'), 22.3 (CH₂, C-6'), 22.2 (CH₂, C-7'); mp:
- 496 177°C; MS (ESI) *m*/z:calcd. for C₁₈H₁₆F₃NO₄S, 399; found 400 [M+H]⁺.
- 497 5-bromo-2-(naphthalene-2-sulfonamido) benzoic acid (11)
- 498 (0.13 g, 67% yield) ¹**H-NMR** (300 MHz, DMSO-d6): $\delta = 10.2$ [s, 1H, COOH], 9.8 [s, 1H, NH] 8.59 [d,
- 499 ${}^{4}J_{1',3'} = 1.4 \text{ Hz}, 1 \text{ H}, 1'-H_{Ar}], 8.17 \text{ [d, } {}^{3}J_{8',7'} = 7.8 \text{ Hz}, 1 \text{ H}, 8'-H_{Ar}], 8.10 \text{ [d, } {}^{3}J_{4'3'} = 8.8 \text{ Hz}, 1 \text{ H}, 4'-H_{Ar}],$
- 500 8.02 [d, ${}^{3}J_{5'.6'} = 7.8$ Hz, 1 H, 5'-H_{Ar}], 7.93 [d, ${}^{4}J_{6.4} = 2.4$ Hz, 1 H, 6-H_{Ar}], 7.77 [dd, ${}^{3}J_{3'.4'} = 8.8$ Hz, ${}^{4}J_{3'.1'}$
- 501 = 1.4Hz, 1 H, 3'- H_{Ar}], 7.72 7.65 [m, 3 H, 4- H_{Ar} , 6'- H_{Ar} , 7'- H_{Ar}], 7.51 [d, ${}^{3}J_{3,4}$ = 8.9 Hz, 1 H, 3- H_{Ar}]. –
- ¹³C-NMR (75 MHz, DMSO-d6): δ = 168.2 (C, C-7) ,138.8 (C, C_{Ar}-2),136.8 (CH, C_{Ar}-4),135.3 (C, C_{Ar}-
- 503 4a'),134.4 (C, C_{Ar}-8a') ,133.4 (CH, C_{Ar}-6) ,131.4 (CH, C_{Ar}-6'), 129.3 (CH, C_{Ar}-4') ,128.5 (CH, C_{Ar}-
- 504 8'),127.8 (2xCH, C_{Ar}-5', C_{Ar}-7') 121.6 (CH, C_{Ar}-3'),120.6 (CH, C_{Ar}-3),119.0 (C, C_{Ar}-1),114.9 (C, C_{Ar}-1)
- 505 5). Mp: 199°C; (ESI) *m*/z: calcd.for C₁₇H₁₁BrNO₄S⁻; 403.9560: found 403.9613 [M-H]⁻.
- 506 5-bromo-2-(phenylmethylsulfonamido)benzoic acid (12)
- 507 (0.07g, 42% yield) ¹**H-NMR** (300 MHz, DMSO-d6): $\delta = 10.57$ [s, 1 H, COO*H*], 8.05 [d, ${}^{4}J_{6.4} = 2.4$
- 508 Hz, 1 H, 6-H_{Ar}], 7.75 [dd, ${}^{3}J_{4,3} = 8.9$ Hz, ${}^{4}J_{4,6} = 2.4$ Hz, 1 H, H-4_{Ar}], 7.49 [d, ${}^{3}J_{3,4} = 8.9$ Hz, 1 H, 3-H_{Ar}],
- $509 \qquad 7.33 7.28 \; [m, 3 \; H, 3' H_{Ar} \; , \; 5' H_{Ar}], \; 7.23 7.20 \; [m, 2 \; H, \; 4' H_{Ar}], \; 5.75 \; [s, 1 \; H, NH], \; 4.72 \; [s, 2 \; H, \; 1' H_{Ar}], \; 5.75 \; [s, 1 \; H, NH], \; 4.72 \; [s, 2 \; H, \; 1' H_{Ar}], \; 5.75 \; [s, 1 \; H, NH], \; 4.72 \; [s, 2 \; H, \; 1' H_{Ar}], \; 5.75 \; [s, 1 \; H, NH], \; 4.72 \; [s, 2 \; H, \; 1' H_{Ar}], \; 5.75 \; [s, 1 \; H, NH], \; 4.72 \; [s, 2 \; H, \; 1' H_{Ar}], \; 5.75 \; [s, 1 \; H, NH], \; 4.72 \; [s, 2 \; H, \; 1' H_{Ar}], \; 5.75 \; [s, 1 \; H, NH], \; 4.72 \; [s, 2 \; H, \; 1' H_{Ar}], \; 5.75 \; [s, 1 \; H, NH], \; 4.72 \; [s, 2 \; H, \; 1' H_{Ar}], \; 5.75 \; [s, 1 \; H, NH], \; 4.72 \; [s, 2 \; H, \; 1' H_{Ar}], \; 5.75 \; [s, 1 \; H, NH], \; 4.72 \; [s, 2 \; H, \; 1' H_{Ar}], \; 5.75 \; [s, 1 \; H, NH], \; 4.72 \; [s, 2 \; H, \; 1' H_{Ar}], \; 5.75 \; [s, 1 \; H, NH], \; 4.72 \; [s, 2 \; H, \; 1' H_{Ar}], \; 5.75 \; [s, 1 \; H, NH], \; 4.72 \; [s, 2 \; H, \; 1' H_{Ar}], \; 5.75 \; [s, 1 \; H, NH], \; 4.72 \; [s, 2 \; H, \; 1' H_{Ar}], \; 5.75 \; [s, 1 \; H, NH], \; 4.72 \; [s, 2 \; H, \; 1' H_{Ar}], \; 5.75 \; [s, 1 \; H, NH], \; 4.72 \; [s, 2 \; H, \; 1' H_{Ar}], \; 5.75 \; [s, 1 \; H, NH], \; 4.72 \; [s, 2 \; H, \; 1' H_{Ar}], \; 5.75 \; [s, 1 \; H, NH], \; 4.72 \; [s, 2 \; H, \; 1' H_{Ar}], \; 5.75 \; [s, 1 \; H, NH], \; 4.75 \; [s, 2 \; H, \; 1' H_{Ar}], \; 5.75 \; [s, 1 \; H, NH], \; 4.75 \; [s, 2 \; H, \; 1' H_{Ar}], \; 5.75 \; [s, 2 \; H, \; 1' H_{Ar}]$
- 510 H] ¹³C-NMR (75 MHz, DMSO-d6): $\delta = 168.3$ (C, C-7), 139.9 (C, C_{Ar}-2), 137(CH, C_{Ar}-4), 133.4 (CH,
- 511 C_{Ar} -6), 130.7 (CH, C_{Ar} -3'), 128.6 (C, C_{Ar} -2'), 128.4 (CH, C_{Ar} -5'), 128.3 (CH, C_{Ar} -4'), 119.5 (CH,
- 512 C_{Ar}-3), 117.5 (C, C_{Ar}-1), 113.9 (C, C_{Ar}-5), 57.4 (CH₂, C-1'). Mp: 216°C; (ESI) m/z: calcd.for
- 513 C₁₄H₁₁BrNO₄S⁻ 367.9860; found 367.9878 [M-H]⁻.
- 5-bromo-2-(4-(phenoxymethyl)phenylsulfonamido)benzoic acid (13)
- 515 (0.6 g, 29% yield) 1H-NMR (300 MHz, DMSO-d6): $\delta = 7.97$ [d, ${}^{4}J_{6,4} = 2.4$ Hz, 1 H, 6-H_{Ar}), 7.85 (d,
- $^{3}J_{2',3'} = 8.3 \text{ Hz}, 2 \text{ H}, 3'-H_{Ar}), 7.73 \text{ [dd, } ^{3}J_{4,3} = 8.9 \text{ Hz}, ^{4}J_{4,6} = 2.4 \text{Hz}, 1 \text{ H}, 4-H_{Ar}], 7.63 \text{ [d, } ^{3}J_{2',3'} = 8.3 \text{ Hz}, 2 \text{ Hz$

- 517 H, 2'-H_{Ar}], 7.47 [d, ${}^{3}J_{3,4} = 8.9$ Hz, 1 H, 3-H_{Ar}], 7.29 [dd, ${}^{3}J_{3",2"} = {}^{3}J_{3",4"} = 7.3$ Hz, 2 H, 3"-H_{Ar}], 7.00 6.92
- 518 [m, 3 H, 4"- H_{Ar} , 2"- H_{Ar}], 5.17 [s, H, 5'-H]. **13C-NMR** (75 MHz, DMSO-d6): δ = 168.2 (C, C-7),
- $519 \quad 157.9 \, (\,C, C_{Ar}\text{-}1") \,, \, 143.2 \, (C, C_{Ar}\text{-}4'), \, 138.8 \, (C, C_{Ar}\text{-}2), \, 137.5 \, (C, C_{Ar}\text{-}1') \,, \, 136.9 \, (CH, C_{Ar}\text{-}4) \, 133.5 \, (CH, C_{Ar}\text{-}1') \,, \, 126.9 \, (CH, C_{Ar}\text{-}1') \,, \, 12$
- 520 C_{Ar} -6), 129.4(CH, C_{Ar} -3"), 128.1(CH, C_{Ar} -2'),127.0 (CH, C_{Ar} -3"), 120.9 (CH, C_{Ar} -4"), 120.5 (CH, C_{Ar} -4"),
- 521 3), 119.0 (C, C_{Ar}-1), 114.9(CH, C_{Ar}-5), 114.7 (CH, C_{Ar}-2"), 68.0 (CH₂, C-5') Mp: 175°C; (ESI) m/z:
- 522 calcd for C20H15BrNO5S- 459.9860 found 459.9878 [M-H]-.

523 5-bromo-2-(2,4,6-trimethylphenylsulfoamido)benzoic acid (14)

- 524 (0.6 g, 78% yield) ¹**H-NMR** (300 MHz, DMSO-d₆): $\delta = 11.77$ [s,1H, COOH], 9.98 [s, 1H, NH], 7.68
- 525 [d, ${}^{3}J_{6,4} = 7.4 \text{ Hz}$, 1H, 6-H_{Ar}], 7.51[dd, ${}^{3}J_{4,3} = 7.1 \text{Hz}$, ${}^{4}J_{4,6} = 7.4 \text{ Hz}$, 1H 4-H_{Ar}], 7.17 [d, 2H, 4'-H_{Ar}, 6'-
- 526 H_{Ar}], 7.14 [d, ${}^{3}J_{3,4}$ = 1H,3- H_{Ar}], 2.56 [s, 6H, CH₃, 9'-H, 7'-H], 2.21 [s, 3H, CH₃, 8'-H];- ${}^{13}C$ -NMR (300)
- 527 MHz, DMSO-d₆): δ = 168.8 (C, C-7), 143.3 (C, C_{Ar}-2), 139.5 (C, C_{Ar}-2'), 139.0 and 139.0 (2xC, C_{Ar}-2')
- 528 3', C_{Ar}-1') 137.3 (CH, C_{Ar}-4), 134.0 (CH, C_{Ar}-6'), 133.0 (CH, C_{Ar}-6), 132.5 and 132.5 (2XCH, C_{Ar}-4',
- 529 C_{Ar}-6') 119.1(CH, C_{Ar}-3), 117.9(C, C_{Ar}-5), 114.3 (C, C_{Ar}-1), 22.5 and 22.5 (2 x CH₃, C-7', C-9') 20.7
- 530 (CH₃, C-8'); mp: 185; MS (ESI): m/z 399 [M+H]+.

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532 2-(4-acetylphenylsulfoamido)-5-(trifluoromethyl)benzoic acid (15)

- 533 (0.4 g, 63% yield) 1 **H-NMR** (300 MHz, DMSO-d6): $\delta = 12.28$ [s, 1H, *COOH*]; 12.10 [s, 1H, *NH*], 8.11
- 534 [d, ${}^{4}J_{6,4}$ = 2.5 Hz, 1H, 6-H_{Ar}], 8.08 [d, ${}^{3}J_{3'2'}$ = 7.5 Hz, 2H, 3'-H_{Ar}], 7.86 [dd, ${}^{4}J_{4,6}$ = 2.5 Hz, ${}^{3}J_{4,3}$ = 7.3Hz,
- 535 1H, 4-H_{Ar}], 7.64 [d, ${}^{3}J_{4,3} = 7.3$ Hz, 1H, 3-H_{Ar}], 7.56 [dd ${}^{3}J_{2',3'} = 7.5$ Hz, ${}^{4}J_{2',6'} = 2.3$ Hz, 2H, 2'-H_{Ar}, 6'-
- 536 H_{Ar}] 7.22 [dd, ${}^{3}J_{3',2'}$ = 7.5Hz, ${}^{4}J_{3',5'}$ = 2.1Hz, 2H, 3'-H_{Ar}, 5'-H_{Ar}] 2.50 [s, 3H, CH₃, 8'-H]; 13 C-NMR
- 538 C_{Ar}-4'), 140.6 (CH, C_{Ar}-4), 131.4 (CH, C_{Ar}-7), 129.6 (2XCH, C_{Ar}-3', C_{Ar}-5'), 128.6 (2xCH, C_{Ar}-2', C_{Ar}-
- 539 6'), 127.6 (C, C_{Ar}-6), 123.0 (C, C-_{Ar}-5), 118.7 (CH, C_{Ar}-3), 27.3 (CH₃, C-8'); mp: 170°C; MS (ESI) *m*/z
- 540 : calcd. for $C_{16}H_{12}F_3NO_5S$. 387; found 388 $[M+H]^+$.

541 2-(2,3-dihydrobenzo[b][1,4]dioxine-6-sulfonamido)-5-(trifluoromethyl)benzoic acid (16)

- 542 (0.4 g, 65% yield) ¹**H-NMR** (300 MHz, DMSO-d6): $\delta = 11.48$ [s, 1H, COOH], 8.13[s, 1H, NH], 7.89
- 543 [d, ${}^{4}J_{6,4} = 3.9 \text{ Hz}$, 1H, 6-H_{Ar}] 7.66 [dd, ${}^{3}J_{4,3} = 7.2 \text{ Hz}$, ${}^{4}J_{4,6} = 4.3 \text{ Hz}$, 1H, 4-H_{Ar}],
- 7.23 [d, ${}^{3}J_{4,3} = 7.2 \text{ Hz 1H, 3-H}_{Ar}$], 7.11 [dd, ${}^{3}J_{2',3'} = 7.3 \text{Hz}$, ${}^{4}J_{2',8'} = 3.2 \text{Hz}$, 1H, 2'-HAr] 6.95 [d, ${}^{4}J_{2',8'} = 3.2 \text{Hz}$]
- 3.2 Hz, 1H, 8'-H_{Ar}] 4.23 4.31 [m, 4H, 5'-H, 6'-H]; 13 C-NMR (75-MHz, DMSO-d6): δ = 168.9(C,

- 546 C-8), 148.3(C, C-4'), 143.8 (C, C-2), 143.5 (C, C-7'), 131.3 (C, C-1'), 130.8 (CH, C-4), 128.6(CH, C-
- 547 6), 125.7 (C, C-7), 122.1 (C, C-5), 120.9 (CH, C-2'), 118.3 (CH, C-3), 118.1 (CH, C-3'), 116.8 (CH, C-10.2)
- 548 8'), 64.7(CH₂, C-5') 64.3 (CH₂, C-6'); mp: 178°C; MS (ESI) *m*/z: calcd. for C₁₆H₁₂F₃NO₆S. 403; found
- 549 404 [M+H]⁺.
- 550 5-(trifluoromethyl)-2-(2,4,6-trimethylphenylsulfoamido)benzoic acid (17)
- 551 (0.38 g, 62% yield) ¹**H-NMR** (300 MHz, DMSO-d6): $\delta = 12.28$ [s, 1H, *COOH*], 11.60 [s, 1H, *NH*], 8.15
- 552 [d, ${}^{4}J_{6,4}$ = 4.3 Hz, 1H, 6-H_{Ar}] 7.92 [dd, ${}^{3}J_{4,3}$ = 7.9 Hz, ${}^{4}J_{4,6}$ = 2.1 Hz, 1H,4-H_{Ar}] 7.87 [d, ${}^{4}J_{6',4'}$ = 1.9 Hz, 2H,
- 553 4'- H_{Ar} , 6'- H_{Ar}], 7.48 [d, ${}^{3}J_{3,4} = 7.9$ Hz, 1H, 3- H_{Ar}], 2.60 [s, 6H, CH₃, 9'-H, 7'-H], 2.23 [s, 3H, CH₃,
- 554 8'-H]; 13 C-NMR (75 MHz, DMSO-d6): δ = 169.3 (C, C-7), 154.2 (C, C-2), 143.6 (C, C-2'), 139.1
- and 139.1 (2xC, C-1', C-3') 132.9 (C, C-5'), 132.5 (CH, C-4), 131.5 and 131.5 (2xCH, C-4', C-6'),
- 556 130.1(CH, C-6), 128.7 (C, C-8), 122.5 (C, C-5), 117.0 (CH, C-3), 109.0 (C, C-1) ,22.4 and 22.4
- 557 (2xCH₃,C-7', C-9'), 20.8 (CH₃, C-8'); mp:184°C; MS (ESI) *m*/z: calcd. for C₁₇H₁₆F₃NO₄S; 387: found
- 558 388 [M+H]⁺.

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562

- 18, 19, 20, and 21 were purchased from Enamine, Kiev, Ukraine as pure compounds (see also Table S6,
- 561 Supporting Information).

563 Determination of ligand binding and binding constant by NMR

- 564 50 μM of ¹⁵N-labeled protein samples were prepared in a 20 mM sodium phosphate buffer containing
- 565 50 mM sodium chloride, 0.02 % (w/v) NaN₃, at pH 7.4. Stock solutions of small molecules were
- prepared in DMSO-d6 at a concentration of 160 mM. A ¹H-¹⁵N HSQC spectrum of Dvl PDZ was
- acquired at 300 K with 5% DMSO-d6 in the absence of ligand as reference spectrum. Mixtures of 16
- 568 compounds were added to ¹⁵N-labeled Dvl PDZ at 8-fold molar excess each. The final concentration of
- 569 DMSO-d6 in the protein-ligand solutions was 5%. Spectra were acquired with 8 scans and 256 points
- in the indirect dimension.
- Binding was deduced if the resonance position of a cross-peak was significantly shifted compared to the
- 572 reference spectrum. The active compound was obtained through successive deconvolution. Experiments
- were recorded on a Bruker DRX600 spectrometer equipped with a triple-resonance cryoprobe. The

preparation of samples was done automatically by a Tecan Genesis RSP 150 pipetting robot. Spectra
 were analysed using the programs TOPSPIN and SPARKY.

Chemical shift perturbations were obtained by comparing the $^{1}\text{H-}^{15}\text{N}$ backbone resonances of protein alone to those of protein-ligand complex. The mean shift difference ($\Delta\delta$ in ppm) was calculated using the equation 1 (Garrett 1997, Bertini 2011).

$$\Delta \delta = \sqrt{\frac{[\Delta \delta_H]^2 + [\Delta \delta_N / 5]^2}{2}} \qquad (Eq. 1)$$

Here $\Delta \delta_N$ and $\Delta \delta_H$ are the amide nitrogen and amide proton chemical shift differences between the free and the bound states of the protein. In order to estimate binding constants, titration experiments monitored by NMR were done. A series of $^1\text{H-}^{15}\text{N}$ HSQC were recorded as a function of ligand concentration. Residues showing a continuous chemical shift change and for which the intensity remained strong were classified as being in fast exchange. The dissociation binding constant was estimated by fitting the observed chemical shift change to equation 2 (Shuker 1996, Hajduk 1997).

586
$$\frac{\Delta\delta}{\Delta\delta_{max}}$$
587
$$= \frac{([L_T] + [P_T] + K_D) - \sqrt{([L_T] + [P_T] + K_D)^2 - 4[L_T] \cdot [P_T]}}{2[P_T]} (Eq. 2)$$

 $\Delta\delta$ is the observed protein amide chemical shift change at a given compound concentration and $\Delta\delta_{max}$ the maximum chemical shift change at saturation. [L_T] the total concentration of the compound, and [P_T] the total concentration of the protein. K_D is the equilibrium dissociation constant. The K_D values are reported as means \pm standard deviations of at least six residues influenced upon binding of the ligand.

Determination of binding constant by Isothermal Titration Calorimetry (ITC)

Isothermal Titration Calorimetry (ITC) experiments were performed using a VP-ITC system (MicroCal). Protein in 20 mM Hepes buffer, 50 mM NaCl, pH 7.4, was centrifuged and degassed before the experiment. A 200 μ M ligand solution containing 2% DMSO was injected 30 times in 10 μ L aliquots at 120 s intervals with a stirring speed of 1000 rpm into a 1.4 mL sample cell containing the Dvl PDZ

domain at a concentration of 20 μ M at 25 °C. Control experiment was initially determined by titrating ligand into buffer at same conditions. Titration of ligand into buffer yielded negligible heats. Thermodynamic properties and binding constants were determined by fitting the data with a nonlinear least-squares routine using a single-site binding model with Origin for ITC v.7.2 (Microcal).

Protein expression

PDZ domains of human AF6 (P55196-2, residues 985–1086) and murine α1-syntrophin (Q61234, residues 81–164) were cloned into pGEX-6P-2 (Amersham Biosciences, Freiburg, Germany) and pGAT2 (European Molecular Biology Laboratory, Heidelberg, Germany), respectively. Proteins were expressed in *E. coli* BL21 (DE3) cells and purified as previously described (Boisguerin 2004). For the cloning of the Dvl-1 PDZ domain (O14640, residues 245–338), IMAGp958J151157Q (ImaGenes) was used as template. V250 is exchanged to isoleucine as in human Dvl-3 or murine Dvl-1. The C-terminal C338 of the domain was exchanged by serine. Via cloning in pET46EK/LIC, a coding sequence for a TEV (Tobacco Etch Virus) protease cleavage site was introduced. The resulting plasmid pDVL1 was transformed in *E. coli* BL21 (DE3). Expression on two-fold M9 minimal medium with 0.5 g/L ¹⁵N NH₄Cl as sole nitrogen source in shaking culture was done at 25 °C overnight with 1 mM IPTG. A yield of 25 mg of pure Dvl-1 was obtained from 1 L culture after IMAC, TEV protease cleavage, a second IMAC, and gel filtration (Superdex 75). The protein domain Dvl-1_245–338 was supplied for NMR in 20 mM phosphate buffer, pH 7.4, 50 mM NaCl.

Crystallization and X-ray diffraction

The His-tagged cleaved human Dvl-3 PDZ domain was concentrated to 12-20 mg/mL in the presence of a 5-fold molar excess of compound 3, 5, 6, 7, 11 and 12. Crystals of all complexes were grown at room temperature by the sitting drop vapour-diffusion method. 200 nL Dvl-3/compound solution was mixed with an equal volume of reservoir solution using the Gryphon (Formulatrix) pipetting robot. Crystals of all complexes were grown to their final size within 4 to 14 days. The Dvl-3 PDZ domain

domains and the 3 PDZ domains of PSD95 was described by Saupe et al (Saupe 2011).

crystallized in complex with compound 3 and 7 in crystallization condition 30% PEG 8000, 0.2 M ammonium sulphate, 0.1 M MES pH 6.5; with compound 5 in 30% PEG 8000, 0.1 M MES pH 5.5; with compound 6 in 1.2 M ammonium sulphate, 0.1 M citric acid pH 5.0; with compound 12 in 32% PEG 8000, 0.2 M ammonium sulphate, 0.1 M Na-cacodylate pH 6.0; with compound 11 in 1 M ammonium sulphate, 1% PEG 3350, 0.1 M Bis-Tris pH 5.5; with compound 12 in 1.26 M sodium phosphate, 0.14 M potassium phosphate and with compound 18 in 1.5 M ammonium sulphate, 12% glycerol, 0.1 M Tris-HCl pH 8.5. The crystals were cryoprotected if necessary, for data collection by soaking for few seconds in precipitant solution containing 20% (v/v) glycerol and subsequently frozen in liquid nitrogen. Diffraction data were collected at 100 K at beamline BL14.1 at the synchrotron-radiation source BESSY, Helmholtz-Zentrum Berlin and processed with XDS.

Structure determination and refinement

Phases for the Dvl-3 PDZ domain in complex with compound 3 were obtained by molecular replacement with PHASER (McCoy 2007) using the *Xenopus laevis* Dishevelled PDZ domain structure (PDB code 2F0A) as a starting model. The reasonable crystal packing and electron density allowed further model and compound building using the program COOT (Emsley 2004) with iterative refinement with REFMAC (Murshudov 1997). All further complex structures were obtained in the same way but using the final refined compound free Dvl-3-PDZ structure as model for molecular replacement. The Ramachandran statistics were analysed by Molprobity (Chen 2010) for all complexes and all crystallographic statistics are given in Supporting Information Tables S2 and S3. Figures were prepared with PyMol. Atomic coordinates and structure factor amplitudes for DVL-3 PDZ domain in complex with compound 3, 5, 6, 7, 11, 12, 18 were deposited in the Protein Data Bank with accession codes 6ZBQ, 6ZBZ, 6ZC3, 6ZC4, 6ZC6, 6ZC7 and 6ZC8, respectively.

MTT assay

- HEK293 cells were plated on a 96-well plate and treated with different concentrations of Dvl inhibitors.
- After 24 h treatment, 20 µl of MTT solution (5 mg/mL) was added into each well. After 2 h incubation,

cell culture medium was replaced with 50 μ L DMSO, and the signal of the purple formazan, produced by living cells, was measured by a plate reader.

TOP-GFP reporter assay

The lentivirus particle (CCS-018L, SABiosciences) encoding GFP under the control of a basal promoter element (TATA box) joined to tandem repeats of a consensus TCF/LEF binding site was transfected into HEK293 cells. Stable cells were selected by puromycin (2 μ g/mL) treatment. Wnt signalling activity indicated by GFP intensity was measured by flow cytometry after 24 h incubation with recombinant mouse Wnt3a (100 ng/mL) or GSK3 inhibitor CHIR99021 (3 μ M) in the presence of Dvl inhibitors.

Luciferase reporter assays

Plasmids encoding a firefly luciferase reporter gene under the control of different responsive elements were transfected into Hela cells with a pRL-SV40 normalization reporter plasmid using the Lipofectamine 2000 (Invitrogen). After desired treatment, cells were harvested in the passive lysis buffer (Promega), and 15 μ L cell lysate were transferred to 96-well LumiNunc plates (Thermo Scientific). Firefly luciferase and Renilla luciferase were detected with the D-luciferin buffer (75 mM Hepes, 4 mM MgSO₄, 20 mM DTT, 100 μ M EDTA, 0.5 mM ATP, 135 μ M Coenzyme A and 100 μ M D-Luciferin sodium salt, pH 8.0) and the coelenterazine buffer (15 mM Na4PPi, 7.5 mM NaAc, 10 mM CDTA, 400 mM Na₂SO₄, 25 μ M APMBT and 1.1 μ M coelenterazine, pH 5.0) respectively using the CentroXS LB960 lumiometer (Berthold Technologies).

Immunoblotting

To assess the β-catenin accumulation in Hela cells, cells were treated with Wnt3a in the presence of Dvl inhibitors for 24 h and lysed in RIPA buffer (50 mM Tris, pH 8.0, 1% NP-40, 0.5% deoxycholate, 0.1% SDS, 150 mM NaCl). Equal amounts of protein were loaded on a SDS-PAGE. Separated proteins were blotted onto PVDF membranes for immunoblot analysis using anti-β-catenin antibody (610154, BD). HRP-conjugated anti-mouse antibody (715-035-150, Jackson ImmunoResearch laboratories) was used

682 for secondary detection with Western lightning chemiluminescence reagent plus (PerkinElmer) and 683 Vilber Lournat imaging system SL-3. 684 685 qRT-PCR analysis 686 To measure the Wnt target accumulation at mRNA level, Hela cells were treated with Wnt3a in the 687 presence of Dvl inhibitors for 24 h. mRNA was extracted according to the standard TRIzol® protocol 688 (Invitrogen) and reverse-transcribed using random primers (Invitrogen) and M-MLV reverse 689 transcriptase (Promega). The qRT-PCR was performed in iQ5 Multicolor Real-Time PCR Detection 690 System (Bio-Rad) using SYBR® Green (Thermo Scientific) and gene-specific primer pairs of Bmp2, 691 Axin2, Lef1 and β -actin (endogenous control). 692 693 Migration assay 694 Cell motility was assessed using 24-well transwell (pore diameter: 8 µm, Corning). SW480WL cells 695 were seeded in the upper chamber in serum free DMEM with 0.1% BSA; 20% serum was supplemented 696 to medium in the lower chamber. After incubation with Wnt3a in the presence of Dvl inhibitors for 24 697 h, nonmigrant cells were scraped off using a cotton swab; the migrated cells on the filters were stained 698 with DAPI, photographed and counted. 699 700 **Colon sphere culture** 701 SW480WL cells were trypsinised into single cells, seeded on 24-well cell culture plates precoated with 702 250 µl polyhema (12 mg/mL in 95% ethanol, Sigma) per well, and incubated with Wnt3a in the presence 703 of Dvl inhibitors in the sphere culture medium (F12: DMEM 1: 1, 1X B-27 supplement, 20 ng/mL 704 EGF, 20 ng/mL FGF, 0.5% methylcellulose) for 10 days. Numbers of spheres were then counted under 705 the microscope. 706

707 Notes

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708 The authors declare no competing financial interest.

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715 ABBREVIATIONS USED

- NMR, nuclear magnetic resonance; HSQC, Heteronuclear Single Quantum Correlation;
- AU, asymmetric unit; SAR, derive structure activity relationships; vdW, van der Waals;
- 718 ITC, Isothermal titration calorimetry; PDZ, PSD95/Disc large/Zonula occludens 1);
- 719 Dvl, Dishevelled; PPI, protein-protein interactions; PDB, Protein Data Bank;
- 720 CSP, chemical shift perturbation; GFP, green fluorescent protein; DMSO, dimethyl sulfoxide;
- PEG, polyethylene glycol; RNA, ribonucleic acid; mRNA, messenger RNA;
- 722 qRT-PCR, quantitative real-time polymerase chain reaction; DMEM Dulbecco's modified Eagle's
- 723 medium; BSA, bovine serum albumin;

725 ASSOCIATED CONTENT

727 Accession Codes

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- 728 Atomic coordinates and structure factor amplitudes for DVL3 PDZ domain in complex with compound
- 3, 5, 6, 7, 11, 12, 18 were deposited in the Protein Data Bank with accession codes 6ZBQ, 6ZBZ, 6ZC3,
- 730 6ZC4, 6ZC6, 6ZC7 and 6ZC8, respectively. Authors will release the atomic coordinates and
- experimental data upon article publication.

733 Supporting information

- 1. Structure-based alignment of the amino acid sequences of Dvl-1,2,3 PDZ; PSD95-PDZ-1,2,3; Af-
- 6 and Syn PDZ domains. (S.2)
- 736 2. 1H-15N HSQC spectra of Dvl-3 PDZ domain alone and in the presence of varying concentrations of
- 737 compound 3. (S.3)

- 738 3. Detailed views of diverse compounds bound to the Dvl-3 PDZ domain. (S.4)
- 739 **4.** Cell viability assays of compounds 3, 7,8, 9, 10, (A) and 18, 20, 21 (B). (S.5)
- 5. ITC binding assays of compound 18 with Dvl-3 PDZ (A) and with Dvl-1 PDZ (B). (S.5)
- 6. Structures of selected compounds used for comparison to our compounds. (S.6)
- 742 7. ITC data of selected compounds used for comparison to our compounds. (S.7)
- 743 **8.** NMR binding assay with compound 322338/3289-8625. (S.8)
- 744 **9.** Purity check of compounds. (S.9)
- 745 Purity check of NPL-1011 compound. (S.9)
- 746 Purity check of Sulindac compound. (S.10)
- 747 Purity check of CalBioChem-322338 compound. (S.11)
- 748 Purity check of NSC668036 compound. (S.12)
- 749 LCMS of intermediate compound 8. (S.13)
- 750 LCMS of intermediate compound 14. (S.13)
- 751 **10.** Chemical shift perturbation values of Dvl-3 PDZ and Dvl-1 PDZ for compounds (3-21). (S.14)
- 752 11. Data collection and refinement statistics of compounds 3, 5, 6, 7. (S.15)
- 753 **12.** Data collection and refinement statistics of compounds 11, 12, 18. (S.16)
- 13. Selectivity of ligands derived from chemical shift perturbation of compounds tested at other PDZ
- 755 domains. (S.17)
- 756 **14.** Details of Multifilter routines. (S.17)
- 757 **15.** Smiles codes and Compounds ID. (S.18)
- 758 **16.** NMR characterization of synthesized compounds (8, 11, 13, 14, 15, 16, 17). (S.21)
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